Syntheses of Dihydropyrene−Cyclophanediene Negative Photochromes Containing Internal Alkenyl and Alkynyl Groups and Comparison of Their Photochemical and Thermochemical Properties

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S Supporting Information

[AB](#page-13-0)STRACT: [Synthesis o](#page-13-0)f a variety of 8,16-disubstituted-anti-[2.2] metacyclophanedienes (CPD) with alkenyl and alkynyl internal (8,16) groups is described together with their analogous dihydropyrenes (DHP). Eyring and Arrhenius parameters were determined for the thermal closing reaction, CPD to DHP, and half-lives at 20 °C were found to range from 11 days ($X = CHO$) to 36 years $(X = CN)$, with alkenyl functions being from 56 days to 10 years. The visible light opening reaction, DHP to CPD, showed relative rates of $1 (X = CN)$ to 240 ($X = CH = CMe_2$).

■ INTRODUCTION

Negative photochromes (−) are not as well-known as their positive $(+)$ counterparts¹ but are interesting because the thermally stable isomer is the more colored one (positive ones have the colorless form, t[h](#page-14-0)e more stable). The colored form bleaches on exposure to visible (longer wavelength) light and returns to the colored isomer on exposure to UV (shorter wavelength) light or in some cases thermally (T). Dihydropyrenes (DHPs), such as 1, are thus negative-thermal $[(-)T]$ photochromes because the deep red-purple 1 opens completely to the colorless cyclophanediene (CPD) form 2 when irradiated with visible light (λ > 400 nm). The latter completely reverts to DHP 1 on irradiation with UV light (λ < 350 nm) and also slowly thermally ($t_{1/2}$ = 7.3 days at 20 °C, 5.75 h at 46 °C) (Scheme 1).²

Scheme 1

Application of photoswitches for use in memory devices requires consideration of their thermal stability, fatigue resistance, overlap of the absorption spectra of the open and closed forms, quantum yield of the interconversions, and ability to nondestructively read the open and closed forms. Our more recent research focus has concentrated on two of these, namely design of more thermally stable dihydropyrene-based photoswitches with high quantum yields of interconversion, particularly for the photoopening reaction of dihydropyrenes with visible light.^{3,4} Although the latter objective is easily addressed by incorporating carbonyl

functionality at the 2 and/or 7 positions of the DHP as in 3, it is at the cost of thermal stability of the CPD 4.5

Since DFT calculations on the transition state of the CPD to DHP interconversion suggested radical character with spin density at the internal $8,16$ positions,⁶ we concentrated on influencing this by first changing the internal group. These calculations suggested that the cyano cy[cl](#page-14-0)ophanediene 5 should have a high activation barrier for thermal cyclization to the dihydropyrene 6. Indeed, it did, with an estimated half-life at room temperature of about 30 years!³ However, a new problem arose in that the corresponding dihydropyrene 6 suffered from an unexpectedly easy (half-life at 50 $^{\circ}$ [C](#page-14-0) = 8.3 h) sigmatropic shift of the cyano groups over the π skeleton of the dihydropyrene to form 7, then 8, and finally 9 by elimination of HCN.

Again we were guided by calculations and subsequently were able to report⁴ that isobutenyl groups at the internal position not only slow down the thermal return of cyclophanediene 10 to dihydropyr[en](#page-14-0)e 11 but also resist the sigmatropic shift of the internal groups over the π system of the dihydropyrene, which result in destruction of the photochrome. In addition, it turned out that the quantum yield of the opening reaction $(11 \text{ to } 10)[\Phi = 0.12]$ was three times higher than that of 1 $[\Phi = 0.039]$.⁴ In that paper,⁴ we reported calculations for a variety of internal groups.

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We now have been able to synthesize several of these, and so this paper reports the synthesis of these molecules and our experimental findings on the photochemical and thermal interconversion reactions.

■ RESULTS AND DISCUSSION

Synthesis. Our first targets were the vinyl- (12), propenyl- $(14, 16, 18)$ and styryl-substituted $(20, 22, 24)$ CPDs and their corresponding DHPs. The route used⁴ was similar to that for the isobutenyl derivative 10, in which conversion of the

Scheme 2

aldehyde groups of 26 was carried out early in the sequence because of the fragility of the dialdehyde 28. Like DHP 6, DHP 28 easily cascaded through a sequence to 31, presumably via 29 and 30.

Thus, Wittig reaction of mixed −SMe isomers of bis-aldehyde $26⁴$ with a preformed ylide derived from methyltriphenylphosphonium bromide and BuLi for 1 h at 20 °C yielded 93% of the de[si](#page-14-0)red bis-vinyl derivative 32 (Scheme 2) as a mixture of isomers. For characterization, a single isomer of 32, in which both −SMe groups are pseudoequatorial, was isolated, mp 228−229 °C. Full characterization for all new compounds is given in the Experimental Section. For synthetic purposes, the mixed isomers of 32 with dimethoxycarbonium tetrafluoroborate¹⁰ gav[e 79% of the bis-sul](#page-5-0)fonium salts 33, which on Hoffmann elimination gave the colorless cyclophanediene 12 in 90% [yiel](#page-14-0)d. In the ${}^{1}H$ NMR spectrum of 12, the protons of the internal vinyl substituent appeared clearly as three doublets of doublets at δ 6.32, 5.47, and 4.84.

Quantitative isomerization of 12 to 13 could easily be obtained either thermally or by irradiation with UV light; however for larger sized samples, it was most easily accomplished thermally. Heating a solution of 12 at 100 °C in toluene under argon for about 1 h was determined to be the optimum time and was easily monitored by following the changes in the ¹H NMR spectrum. In 13, the DHP external protons appearing at δ 8.79 (s), 8.65 (d), and 8.08 (t) are significantly deshielded from the analogous ones of 12 by 2.41, 2.05, and 1.01 ppm, respectively. The internal vinyl protons at δ 2.67, 2.06, and 0.50 (all as doublets of doublets) are highly shielded by 2.17, 3.41, and 5.82 ppm, respectively, caused by the strong DHP ring current from those quoted above for 12.

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Wittig reaction of the bis-aldehyde 26 with the ylide derived from ethyltriphenylphosphosphonium bromide and BuLi, at 20 °C for 1 h, gave a good yield of 34 as a complex mixture of cyclophane−SMe and double bond (ZZ)-, (ZE)-, and (EE)-isomers (Scheme 3), and so the mixture was directly

converted by Hoffmann elimination as for 12 to a 6:1 mixture of 14 (ZZ) and 16 (ZE) . Existence of the third isomer 18 (EE) in this mixture was not confirmed.

The (ZZ)- or cis−cis-isomer 14 was expected (nonstabilized ylide) to predominate, and did, and in its $^1\mathrm{H}$ NMR spectrum, the cis orientation of the double bond was established by a coupling constant of 12.0 Hz between the olefinic protons and by an IR band at 711 cm[−]¹ . All protons and carbons in the major isomer 14 could easily be accounted for but in the minor isomer 16, the cis-propenyl protons and carbons were obscured by overlap with peaks from the major isomer. The transpropenyl internal group in 16 was characterized by a coupling constant of 15.4 Hz for the alkene proton at δ 5.94 and by an IR band at 962 cm⁻¹. DHPs 15 and 17 were obtained preparatively by thermal isomerization of 14 and 16, as above, or by irradiation with UV light (λ_{max} < 300 nm). The characteristic

Scheme 4

¹H NMR peaks for the dihydropyrene skeleton were found in the region of δ 8−9, while the internal olefinic (δ 3.21 and 0.09) and methyl protons $(\delta 0.33(15)$ and 0.32 and 0.17 (17)) were quite shielded, being in the center of the ring current.

The styryl-substituted CPDs were obtained analogously (see Scheme 4) as a mixture of 20 (ZZ, 44%) 22 (EZ, 38%), and 24 (EE, 18%) that were easily separated by column chromatography. The styryl olefinic and aromatic protons were quite distinct for the (Z) - and (E) -isomers. The olefinic protons of the (Z) -styryl group in 20 appeared at δ 5.96 and 5.80 whereas in the (E) -styryl fragment in 24 these protons appeared accidentally as a singlet at δ 6.86.

Thermal isomerization of the (ZZ) -styryl CPD 20 (as described above) gave (ZZ)-styryl DHP 21 as orange crystals from cyclohexane, mp 152−154 °C. The olefinic protons in 21 appeared shielded at δ 0.28 and 4.21 as doublets (*J* = 12.6 Hz). The ortho protons of the styryl aromatic rings are also shielded to a small extent by the DHP ring current. The (EE)-styryl DHP 25 was obtained as dark green crystals, mp 232−234 °C. A significant difference in λ_{max} is observed between 21 and 25. The absorption wavelength in the visible region for 25 is observed at 466 and 597 nm whereas the corresponding bands for 21 are observed at 487 and 624 nm. The olefinic protons in 25 appeared at δ 0.87 and 3.21.

To explore the electronic effect of substituents on the thermal behavior of cyclophanedienes, functionalized styryl substituents were also introduced. We synthesized the $−OCH₃$, −NO2, and −CH3 substituted styryl CPDs 43−51 as representative examples of electron -donor, -acceptor, and -neutral (hyperconjugation) substituents. A general synthesis of these cyclophanedienes and then conversion into dihydropyrenes is shown in Scheme 4; however, details of the syntheses, yields,

ratios of isomers, and characterizations are given in the Experimental Section.

As examples with extended linear conjugation, cyclophanediene 61 [represents a](#page-5-0) simple example; however, all attempts to synthesize this compound resulted in a nonseparable mixture of isomers. On the other hand, the methyl-substituted analogues 62−64 could be easily synthesized as a mixture of (ZZ):(ZE): (EE) isomers which could be separated by column chromatography. Syntheses of compounds 62−67 were analogous to those of 20−25 above and are described in detail in the Experimental Section.

In order to investigate an acetylene-substituted CPD, the Wittig reaction of bis-aldehyde cyclophane 22 with the ylide derived from bromomethyltriphenylphosphonium bromide and KOtBu at 20 °C for 1 h with subsequent in situ elimination of HBr yielded the bis-acetylene cyclophane 68 as a mixture of isomers which was directly subjected to a sequence of S-methylation with $(CH_3O)_2CHBF_4$ and Hoffmann elimination as previously described to give the bis-acetylene CPD 69 as colorless solid. The alkyne was characterized by an IR \equiv C−H stretch at 3293 cm⁻¹. Bis-acetylene DHP 70 was obtained by irradiation of cyclophanediene 69 with UV light as dark green crystals, with the internal alkyne protons appearing at δ –0.09, and the IR \equiv C−H stretch at 3274 cm⁻¹. Note that neither 69 or 70 showed a clear $C\equiv C$ stretch.

For comparison purposes, the unsymmetrical cyclophanedienes 77 and 84 were also synthesized, starting from 71^{11} or 79 with $72₁¹²$ as shown in Scheme 5, with details in the Experimental Section. Dibromide 79 required for the syn[the](#page-14-0)sis of 84 was s[yn](#page-14-0)thesized in the three-st[ep](#page-4-0) process¹³ shown in [Scheme 6 from comm](#page-5-0)ercially available 2,6-dimethylaniline 86.

Thermochemical Results. The objective of [our](#page-14-0) work was to pro[du](#page-4-0)ce cyclophanedienes which have relatively slow thermal return reactions to the corresponding dihydropyrenes. Following the thermal return reaction is relatively straightforward using either UV/vis spectroscopy^{5b} or by ¹H NMR spectrometry^{3,5c} since the CPDs show very different spectra from the DHPs. Then use of Arrhenius pl[ots](#page-14-0) leads to E_{act} and ln A data, whil[e Ey](#page-14-0)ring plots yield ΔH^{\ddagger} and ΔS^{\ddagger} data. However, understanding the thermal cyclization of CPDs to DHPs just by looking at the thermochemical data proves to be more difficult than one might imagine; 7 for example, direct comparison of energies of activation (E_{act}) tends to be somewhat misleading for this reaction because t[h](#page-14-0)ere appears to be a large variation in the pre-exponential factor $(\ln A)$ from system to system.³ We have thus found that comparison of thermal conversion

half-lives $(t_{1/2})$ at two or three temperatures to be more useful. While the full Eyring and Arrhenius data are given in the Supporting Information, Table 1 gives the half-life data determined for the compounds in this study.

The observed half-life values follow surprisingly well the calculated activation barriers:⁴ for example, both ΔH^{\ddagger} and $\tau_{1/2}$ (20 °C) follow the order $10 > 14 > 20$. The (ZZ)-internal alkenes thermally convert m[or](#page-14-0)e slowly than the (ZE)-isomers, which are slower than the (*EE*)-isomers, e.g., for $\tau_{1/2}$ (20 °C) $20 > 22 > 24$. Para-substituents on the phenyl ring of 20 seem to have a larger effect on the (ZZ)-isomer than on the others. The mixed internal group compound 77 appears to have a barrier approximately midway between 5 and 89. It is worth noting that our initial choice for target, 4 the bis-isobutenyl compound 10, was a good one, as all of the compounds that follow have faster thermal conversions, t[h](#page-14-0)ough all are much slower and thus are better than the parents 89 or 90 .

Photo-opening Ractions. Visible light induced relative opening rate studies were performed using a 500 W tungsten lamp with an orange plastic ∼490 nm cutoff filter (see the Supporting Information for spectrum) using ¹H NMR studies in toluene- d_8 , side by side with the standard, as we have [described previously.](#page-13-0)^{5c} Because the compounds tested in this study are slower opening than benzo-DHP 1, first the styryl compound 52 was m[ea](#page-14-0)sured against 1 and then 52 was used to calibrate the others that are given in Table 2. Our initial target, the isobutenyl DHP 11, still proved to open fastest, at about 25% [o](#page-5-0)f the rate of 1, but the (ZZ) -isomers of the styryl, p -Me, and p -OMe styryl compounds $(21, 52,$ and $55)$ all showed very similar and acceptable rates and all opened completely. Some of the compounds (indicated in Table 2 by the *) formed photostationary states. Surprisingly, the (EE)-styryl compounds 25, 54, 57, and 60, the (EE)-butadienyl [co](#page-5-0)mpound 67, and the phenylethynyl compound 85 did not result in any measurable open (CPD) form under these conditions, though it is possible that the CPD tail extended far enough into the visible to permit any "open" form to close again (the quantum yield of the photoclosing reaction is close to 1, much higher than that of the photoopening reaction⁹). Certainly when the trans−trans-styryl CPD 24 was irradiated under the same conditions, it closed to the DHP 25. Howeve[r,](#page-14-0) use of a 590 nm red plastic filter also failed to open it. The (ZZ) - and (ZE) -butadienyl compounds 65 and 66 only isomerized to the (EE)-isomer 67. The vinyl compound 13 did not open appreciably, but decomposed. Robb⁸ has shown that the photoopening reaction of the parent 91 occurs through a conical intersection between a biradical excit[ed](#page-14-0) state and the ground one. However, the former is not the lowest energy excited state, and moreover excitation to it is symmetry forbidden. Stabilization of this state relative to the highly populated zwitterionic excited state should accelerate the photoopening reaction. However, it is not obvious whether this would be a dominating factor or not and how substituents affect this. Compound 15 with cis-propenyl groups completely

Scheme 5

Scheme 6

 a^a h = hour[s;](#page-14-0) d = days; m = minutes. Errors estimated are <5%.

Table 2. Visible Light Opening Rates for the Dihydropyrenes to the Cyclophanedienes, Measured Relative to that of 1 = 1000 with Calculated Rates Relative to Parent 91

| compd (int group) | rates vs DHP $91 = 1000$ | calcd ^b relative relative rates ^c vs benzo DHP $1 = 1000$ | $CPD: DHP^d$ |
|---|-----------------------------|---|--------------|
| 6^a –CN | \sim 6 | \sim 1 | 60:40 |
| 70^a –C \equiv CH | 65 | 4 | 60:40 |
| 15 (ZZ) -CH=CHMe | 180 | 11 | |
| 58^a (ZE) $-CH = CHPh$ $NO2-p$ | 200 | 12 | 20:80 |
| $23a$ (ZE) $-CH=CHPh$ | 740 | 45 | 80:20 |
| $53a$ (ZE) -CH=CHPh- $Me-p$ | 980 | 60 | 50:50 |
| 56^a (ZE) $-CH = CHPh$ $OMe-p$ | 1130 | 69 | 60:40 |
| $21 (ZZ)$ -CH=CHPh | 2500 | 150 | |
| 55 (ZZ) -CH=CHPh- $OMe-p$ | 2900 | 180 | |
| $52 (ZZ)$ -CH=CHPh- $Me-p$ | 2900 | 180 | |
| $11 - CH = CMe$ | 3900 | 240 | |
| ${}^{a}P$ = photostationary state. ^b Approximately 16.25X rate for 1. ^{5c c} Error estimates are less than 5%. d Estimated by ¹ H NMR. | | | |

opened, while 17, with one trans-group did not open at all! For the styryl compounds, the (ZE)-compounds do open, though more slowly than the (ZZ)-isomers. A qualitative explanation can be made on examination of the UV−vis spectra of the styryl-CPDs shown in Figure 1 (equimolar concentrations).

Figure 1. Comparison of the UV−vis spectra tails for the styryl CPDs 20, 22, and 24.

For *cis,cis-20* (blue, Figure 1), the tail has reached baseline before 510 nm. For cis,trans-22 and trans,trans-24 the tails show increasingly significant tails above 500 nm, providing a qualitative explanation for DHP, cis,cis-21 having the fastest opening rate, while for the trans compounds the photoclosing reaction increasingly competes.

For the para-substituted cis,cis-styryl CPDs shown in Figure 2 (equimolar concentrations), again examination of the CPD tails

Relative absorbancies of the UV-vis tails of selective cis styryl CPDs

Figure 2. Comparison of the UV−vis spectra tails for the parasubstituted cis,cis-styryl CPDs 20, 43, 46, and 49.

is consistent with the p -nitro DHP 58 showing the slowest rate of opening with considerable CPD content of the photostationary state, while the p -Me DHPs 52 shows the fastest rate and complete conversion.

Sigmatropic Rearrangement. The remaining objective that we have not yet discussed is the sigmatropic rearrangement that occurs >50 °C for the dicyano DHP 6 which ultimately led to 9.³ Happily, for the bis-isobutenyl DHP 11, the rearrangement is suppressed and is not observed at all at 100 °C. With the [ex](#page-14-0)ception of the diformyl DHP 28 which isomerized to 29 (E_{act}) estimated to be <20 kcal/mol) and then on to 31 even more easily than for 6, the other DHPs prepared in this study all showed no evidence of rearrangement at 100 °C.

■ **CONCLUSIONS**

In this paper, the synthesis of a number of cyclophanedienes with a variety of different internal alkenyl and alkynyl groups and their corresponding dihydropyrenes is described. The thermal conversions of the CPDs to DHPs and the photochemical conversions of the DHPs to CPDs are measured and compared. CPDs with cis-alkenyl internal groups showed the slowest thermal conversion to DHPs, with half-lives of several years at 20 °C, which is much better than that observed for benzo CPD 2, but not as good as for isobutenyl CPD 10. The photoopening reaction rate for the corresponding *cis*alkenyl DHPs was similar to that of DHP 11. As well, for the compounds studied, the sigmatropic rearrangement was suppressed. Of the compounds studied here, the CPD/DHP pairs $20/21$, $43/52$, and $46/55$ show the most promising properties as photochromic switches. The relative observed thermal closing rates, CPD to DHP, agree reasonably well with previous⁴ calculations.

EX[PE](#page-14-0)RIMENTAL SECTION

The general conditions and the numbering system used for NMR assignments are given in the Supporting Information.

General Thermolysis Procedure Used for Converting CPDs **into DHPs.** The cyclophanediene (25 mg) in toluene- d_8 (2 mL) was sealed in an NMR tube under [argon, which was then h](#page-13-0)eated at 100 °C until ¹H NMR indicated conversion to the dihydropyrene was complete (∼1−2 h). Evaporation yielded the DHP as colored crystals. 2,10-Bis(methylthio)-8,16-diethenyl-anti-[2.2]metacyclophane (32). BuLi (5.3 mL, 2.5 M in hexanes, 13 mmol) was added dropwise to a stirred suspension of methyltriphenylphosphonium bromide

(3.6 g, 10.1 mmol) under argon in THF (40 mL) at 0 °C. After 20 min,

diformylmethylthiocyclophane $26⁴$ (1.5 g, 4.21 mmol) was added to this clear orange-red solution, and stirring was continued at 20 °C for 1.5 h. Water was then added [fo](#page-14-0)llowed by dichloromethane. The organic extract was washed, dried, and evaporated. The crude product was chromatographed over silica gel using hexanes/dichloromethane (7:3) as eluent and gave 1.38 g (93%) of divinylmethylthiocyclophane 32 as a mixture of isomers: EIMS m/z 352 (M⁺); ¹H NMR δ 7.98 (d, J = 7.6 Hz), 7.80 (d, J = 7.7 Hz), 7.67−7.62 (m), 7.53−7.51 (m), 7.45−7.42 (m), 7.20−7.16 (m), 7.08−7.05 (m), 6.94 (t, J = 7.4 Hz), 4.80−4.60 (m), 4.52−4.40 (m), 4.28−4.25 (m), 4.11−4.08 (m), 3.08− 3.04 (m), 2.87 (t, $J = 12.0$ Hz), 2.66 (t, $J = 12.1$ Hz), 2.12, 2.08 (s). For synthetic purposes, these were used directly in the next step. For characterization purposes, a single isomer of 32, in which the 1,9 methylthio groups are pseudoequatorial was obtained by rechromatography as colorless crystals: mp 228−229 °C; ¹H NMR δ 7.81 (dd, $J = 7.6$, 1.1 Hz, 2H, H-6,14), 7.18 (dd, $J = 7.5$, 1.0 Hz, 2H, H-4,12), 7.07 (t, J = 7.5 Hz, 2H, H-5,13), 4.80−4.75 and 4.48−4.42 (m, 6H, H-19,20,17,18), 4.27 (dd, J = 11.6, 4.2 Hz, 2H, H-1,9), 3.06 (dd, J = 12.6, 4.2 Hz, 2H, H-2_{eq},10_{eq}), 2.66 (dd, J = 12.1, 11.9 Hz, 2H, H-2_{ax},10_{ax}), 2.13 (s, 3H, SMe); ¹³C NMR δ 147.0 (C-8,16), 135.7 (C-7,15), 135.2 $(C-3,11)$, 135.0 $(C-17, 19)$, 128.9 $(C-4,12)$, 126.6 $(C-5,13)$, 125.4 $(C-6,14)$, 120.6 $(C-18,20)$, 53.2 $(C-1,9)$, 43.3 $(C-2,10)$, 15.3 $(C-$ 21,22); IR ν (KBr) 3078, 2960, 2909, 1438, 1397, 994, 930, 771, 748 cm^{−1}; UV−vis (cyclohexane) λ_{max} nm (ε_{max}) 227 (17300), 288 (3986); EIMS m/z 352 (M⁺); HRMS calcd for $C_{22}H_{24}S_2$ 352.1319, found 352.1319. Anal. Calcd: C, 74.95; H, 6.86. Found: C, 74.48; H, 6.91.

1,10-Bis(methylthio)-8,16-divinyl-anti-[2.2]metacyclophane Bissulfonium Salt (33). The mixed isomers of 32 (360 mg, 1 mmol) in dry CH_2Cl_2 (20 mL) were added slowly to $(MeO)_2CHBF_4$ (Borch reagent)¹⁰ (80% oil, 580 mg, 2.86 mmol) in CH_2Cl_2 (14 mL) at −78 °C with stirring under nitrogen. The mixture was then stirred at 20 °C f[or](#page-14-0) 3 h. The CH₂Cl₂ was then decanted from the oil, ethyl acetate (40 mL) was added, and stirring was continued for another 3 h (this was repeated if a nonsticky white powder was not obtained). The white precipitate was collected and dried to give 440 mg (79%) of the sulfonium salt 33: ¹H NMR (DMSO- d_6) δ 7.72 (d, J = 7.8 Hz), 7.58 $(d, J = 7.4 \text{ Hz})$, 7.54 $(d, J = 7.3 \text{ Hz})$, 7.41 $(d, J = 7.6 \text{ Hz})$, 7.28 $(t, J =$ 7.6 Hz), 7.13 (t, J = 7.4 Hz), 5.2−4.6 (m), 3.46−3.05 (m), 2.90, $2.87(s)$. This was used directly in the next step.

8,16-Divinyl-anti-[2.2]metacyclophane-1,9-diene (12). t-BuOK (250 mg, 2.23 mmol) was added to a stirred suspension of the bissulfonium salt 33 (440 mg, 0.791 mmol) in THF (20 mL) under argon at 20 °C in a vessel wrapped in aluminum foil to exclude light. After the mixture was stirred for 30 min, water and CH_2Cl_2 were added. The organic extract was washed, dried, and evaporated (all with protection from light). The product was chromatographed over silica gel using hexanes as eluent gave 182 mg (90%) of 12 as colorless crystals from cyclohexane: ${}^{1}H$ NMR δ 7.07 (t, J = 7.4 Hz, 2H, H-5,13), 6.60 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.38 (s, 4H, H-1,2,9,10), 6.32 $(dd, J = 17.4, 11.0 Hz, 2H, H-17,19)$ 5.47 $(dd, J = 17.4, 1.8 Hz, 2H, H-17$ 18,20b), 4.84 (dd, J = 11.0, 1.8 Hz, 2H, H-18,20a); ¹³C NMR δ 146.4 $(C-8,16)$, 135.8 $(3,7,11,15)$, 135.2 $(C-17,19)$, 133.2 $(C-1,2,9,10)$, 129.5 (C-5,13), 126.7 (C-4,6,12,14), 118.9 (C-18,20); IR ν (KBr) 3077, 3049, 3004, 1618, 1436, 1397, 993, 903, 861, 807, 764, 605 cm⁻¹; UV−vis (cyclohexane) λ_{max} nm (ε_{max}) 202 (51700), 242 (33170), 279 (12370), 339 (11000), 377 (5710), 390 sh (2100); EIMS m/z 256 (M⁺); HRMS calcd for $C_{20}H_{16}$ 256.1252, found 256.1255. Anal Calcd: C, 93.70; H, 6.30. Found: C, 93.51; H, 6.41. Attempted mp determination isomerized 12 into DHP 13.

10b,10c-Divinyl-trans-10b,10c-dihydropyrene (13). (a) Thermolysis (Best for Preparative Samples). Cyclophanediene 12 (25 mg) in toluene- d_8 (2 mL) was sealed in an NMR tube under argon, which was then heated at 100 $^{\circ}$ C until ¹H NMR indicated conversion to dihydropyrene 13 was complete (∼1−2 h). Evaporation yielded 25 mg (100%) of 13 as dark green crystals from cyclohexane: mp 163−165 $^{\circ}$ C; ¹H NMR δ 8.79 (s, 4H, H-4,5,9,10), 8.65 (d, J = 7.6 Hz, 4H, H-1,3,6,8), 8.08 (t, J = 7.6 Hz, 2H, H-2,7), 2.67 (dd, J = 10.3, 1.4 Hz, 2H, H-12,14b), 2.06 (dd, J = 17.0, 1.5 Hz, 2H, H-12,14a), 0.50 (dd, J = 17.0, 10.3 Hz, 2H, H-11,13); 13C NMR δ 133.8 (3a,5a,10a,10d), 128.6 (C-11,13), 125.5 (C-4,5,9,10), 124.4 (C-1,3,6,8), 123.8 (C-2,7), 109.8

(C-12,14), 36.1 (C-10b,10c); IR ν (KBr) 3032, 3007, 1621, 1399, 1295, 912, 841 cm⁻¹; UV-vis (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 339 (71300), 377 (30000), 462 (5600), 608 (100); EIMS m/z 256 (M⁺); HRMS calcd for $C_{20}H_{16}$ 256.1252, found 256.1251. Anal. Calcd: C, 93.70; H, 6.30. Found: C, 93.92; H, 6.57.

(b) UV Irradiation (Only Used for Small Samples To Indicate That the Reaction Proceeds) . The cylophanediene 12 (2 mg) was dissolved in sufficient cyclohexane or dichloromethane to fill a quartz UV cuvette, which was then irradiated with 254 nm light from a handheld TLC lamp. The irradiation was monitored at 15 s intervals until complete conversion to dihydropyrene 13 was observed (the bands at 462 and 608 nm maximize). This required 45−75 s. The UV−vis spectrum observed was the same as that obtained for thermolysis, method a.

8,16-Bis(prop-1-enyl)-anti-[2.2]metacyclophane-1,9-dienes (14) and (16). Using the same procedure as for the sequence 26−32− 33−12 above, except that ethyltriphenylphosphonium bromide was used, followed by the same methylation/Hoffmann elimination sequence, yielded after chromatography 202 mg (90%) of the two isomers 14 and 16 in approximately a ratio of 6:1. By further chromatography, these could be partially separated. One fraction was enriched in the major isomer, (ZZ) -14: ¹H NMR δ 6.94 (t, J = 7.4 Hz, 2H, H-5,13), 6.46 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.12 (qd, J = 12.0, 1.8 Hz, 2H, H-17,19), 6.09 (s, 4H, H-1,2,9,10), 5.23 (qd, J = 12.0, 7.2 Hz, 2H, H-18,20), 1.55 (dd, J = 7.3, 1.8 Hz, 6H, H-21,22); ¹³C NMR δ 142.3 (C-8,16), 136.9 (C-3,7,11,15), 132.0 (C-1,2,9,10), 129.4 (C-17,19), 128.2 (C-5,13), 126.0 (C-18,20), 125.5 (C-4,6,12,14), 15.4 (C-21,22); IR ν (thin film) 3036, 3001, 2930, 2848, 1599, 1561, 1434, 962, 929, 858, 842, 771, 711, 665 cm⁻¹; UV–vis (cyclohexane) λ_{\max} nm (ε_{\max}) 215 (25000), 243 (26200), 277 (11500), 340 (∼ 2500); EIMS m/z 284 (M⁺); HRMS calcd for $C_{22}H_{20}$ 284.1565, found 284.1562.

The second fraction was enriched in the minor isomer, (ZE) -16: ^{1}H NMR δ 6.98 (t, J = 7.4 Hz, 1H, H-13), 6.86 (t, J = 7.4 Hz, 1H, H-5), 6.44 (d, J = 7.4 Hz, 2H, H-12,14), 6.42 (d, J = 7.4 Hz, 2H, H-4,6), 6.10−6.14 (AB*, 4H, H-1,2,9,10), 5.94 (dq, J = 15.4, 6.7 Hz, H-20), 1.57 (dd, $J = 1.9$ Hz, $J = 6.6$ Hz, 3H, H-21), 1.52 (dd, $J = 6.6$, 1.6 Hz, 3H, H-22) H-17−19*; 13C NMR δ 145.9 (C-8), 141.3 (C-16), 137.4 (C-11,15), 135.1 (C-3,7), 132.6 (C-2,9), 131.8 (C-17), 130.8 (C-19), 129.8 (C-20), 128.5 (C-13), 127.8 (C-5), 126.2 (C-4,6), 125.4 (C-12,14), 18.2 (C-22), 15.3 (C-21), C-1,10,18*. *Indicates when peaks are not distinct (overlapped by the major isomer).

10b,10c-Bis(prop-1-enyl)-trans-10b,10c-dihydropyrenes (15) and (17). Using the same thermolysis procedure used for 12 above, CPD 15 (25 mg) was converted to (ZZ) -DHP 15 (25 mg, 100%): ¹H NMR δ 8.64 (d, J = 7.6 Hz, 4H, H-1,3,6,8), 8.63 (s, 4H, H-4,5,9,10), 8.05 (t, $J = 7.6$ Hz, 2H, H-2,7), 3.21 (qd, $J = 12.4$, 7.3 Hz, 2H, H-12, 14), 0.33 $(dd, J = 7.3, 1.8$ Hz, 6H, H-15,16), 0.09 (qd, J = 12.4, 1.8 Hz, 2H, H-11,13); 13C NMR δ 135.3 (C-3a,5a,10a,10d), 126.1 (C-4,5,9,10), 125.4 (C-1,3,6,8), 123.61 (C-12,14), 123.59 (C-2,7), 121.9 (C-11,13), 11.5 (C-15,16); EIMS m/z 284 (M⁺); HRMS calcd for C₂₂H₂₀ 284.1565, found 284.1557.

Thermolysis of the mostly (ZE)-isomer 16 gave 100% of (ZE)- DHP 17: ¹H NMR δ 8.71 (d, J = 7.7 Hz, 2H, H-4,10), 8.67 (d, J = 7.6 Hz, 2H, H-6,8), 8.66 (d, J = 7.6 Hz, 2H, H-5,9), 8.58 (d, J = 7.7 Hz, 2H, H-1,3), 8.07 (t, J = 7.8 Hz, 1H, H-7), 8.02 (t, J = 7.7 Hz, 1H, H-2), 3.14 (qd, J = 12.4, 7.3 Hz, 1H, H-14), 2.36 (qd, J = 15.1, 6.5 Hz, 1H, H-12), 0.32 (dd, J = 7.2, 1.8 Hz, 3H, H-15), 0.17 (dd, J = 6.7, 1.5 Hz, 3H, H-16), 0.15 (m, 2H, H-11,13); ¹³C NMR δ 135.2 (C-5a,10d), 134.8 (C-10a,3a), 126.1 (C-5,9), 125.2 (C-4,10), 125.1 (C-6,8), 124.2 (C-1,3), 123.44 (C-2), 123.39 (C-7), 123.0 (C-13), 122.8 (C-14), 120.6 (C-11), 120.3 (C-12), 16.4 (C-15), 11.1 (C-16); IR ν (KBr) 3018, 2923, 2852, 1365, 836, 797, 692 cm[−]¹ ; UV−vis (cyclohexane) λ_{max} nm (ε_{max}) 346 nm (46600), 369 (12700), 390 (21100), 482 (4500), 626 (120). Anal. Calcd: C, 92.91; H, 7.09. Found: C, 92.91; H, 7.10.

8,16-Bis(2-phenylethenyl)-1,9-bis(methylthio)-anti-[2.2] metacyclophane (35). Using the same procedure as for 22 to 32 above, diformylmethylthiocyclophane 22 (1.40 g, 3.93 mmol) with the ylide prepared by reaction of t-BuOK (4.0 g, 36 mmol) and benzyl triphenylphosphonium bromide (5.0 g, 11.5 mmol) in THF (30 mL)

followed by column chromatography using hexanes/dichloromethane (7:3) gave 400 mg (20%) of mixed isomers of distyrylmethylthiocyclophane 35: ¹H NMR δ 7.98 (d, J = 7.7 Hz), 7.96 (d, J = 8.0 Hz), 7.91 (d, J = 7.6 Hz), 7.84 (d, J = 7.6 Hz), 7.38−6.90 (m), 6.34−6.27 (m) , 5.93 (d, J = 12.4 Hz), 5.88 (d, J = 12.4 Hz), 5.73 (d, J = 16.4 Hz), 5.22 (d, J = 16.4 Hz), 5.16 (d, J = 16.4 Hz) 4.42–4.00 (m), 3.16–2.95 (m), 2.73−2.53 (m), 2.19, 2.18 (s), 1.61, 1.38, 1.37 (3 singlets); EIMS m/z 504 (M+); HRMS calcd for $C_{34}H_{32}S_2$ 504.1945, found 504.1938. These were used directly in the next step.

8,16-Bis(2-phenylethenyl)-1,9-bis(methylthio)-anti-[2.2] metacyclophane Bis-sulfonium Salt (39). Using the same procedure as for 33 above, from the mixed isomers of 35 (600 mg, 1.19 mmol) with Borch reagent (80% oil, 1g, 5 mmol) in CH_2Cl_2 (10 mL) gave 700 mg (83%) of bis-sulfonium salt 39: ¹H NMR (DMSO- d_6) δ 7.75− 7.18 (m), 7.08–6.96 (m), 6.37–5.91 (m), 5.33 (d, J = 16.4 Hz), 4.61– 4.51 (m), 4.28−3.95 (m), 3.48−3.37 (m), 3.29−3.15 (singlets) 2.98− 2.80 (singlets), 2.27, 2.29 (s), 2.14, 1.96 (s), which were used directly in the next step.

8,16-Bis(2-phenylethenyl)-anti-[2.2]metacyclophane-1,9-dienes (20), (22), and (24). Using the same procedure as for 12 above, the bis-sulfonium salt 39 (650 mg, 0.93 mmol) and t-BuOK (500 mg, 4.5 mmol) in THF (10 mL) gave a mixture of cyclophanedienes which was purified by column chromatography using hexanes-dichloromethane 85: 15 as eluent. Elued first was the cis−cis or bis-(1Z) isomer 20, 152 mg (41%), as pale yellow crystals from cyclohexane. ¹H NMR (CD₂Cl₂) δ 7.18 (t, J = 7.4 Hz, 2H, H-5,13), 7.09–7.02 (m, 6H, H-27−32), 6.81−6.77 (m, 4H, 23−26), 6.59 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 5.96 (d, J = 12.4 Hz, 2H, H-19,20), 5.84 (s, 4H, 1,2,9,10), 5.80 (d, J = 12.4 Hz, 2H, H-17,18); ¹³C NMR (CD₂Cl₂) δ 141.9 (C-8,16), 139.1 (C-21,22), 138.6 (3,7,11,15), 132.7 (C-1,2,9,10), 131.6 (C-19,20), 130.2 (C-5,13), 123.0 (C-17,18), 128.8 (C-23−26), 128.2 (C-31,32), 126.8 (C-27−30), 126.6 (C-4,6,12,14); IR ν (KBr) 3077, 3050, 3005, 1596, 1492, 1448, 1432, 961, 795, 769, 736, 702, 696, 646, 597 cm⁻¹; UV–vis (cyclohexane) λ_{\max} nm (ε_{\max}) 257 (28900), 280 (23900), 389 (3700); EIMS m/z 408 (M⁺); HRMS calcd for $C_{32}H_{24}$ 408.1878, found 408.1884. On attempted melting point determination CPD 20 converted into DHP 21.

Eluted second was the *cis–trans* or $8-(1E)$,16-(1Z) isomer 22 as 132 mg (35.4%) of pale yellow crystals from cyclohexane: ¹H NMR δ 7.24−7.00 (m, 8H, 23,24, 27−32), 7.13 (t, J = 7.4 Hz, 1H, H-13), 6.83−6.80 (m, 2H, H-25,26), 7.00 (t, J = 7.4 Hz, 1H, H-5), 6.81 and 6.73 (AB, $J = 16.3$ Hz, H-17,19), 6.72 (d, $J = 7.4$ Hz, 2H, H-12,14), 6.45 (d, J = 7.4 Hz, 2H, H-4,6), 6.33 (d, J = 11.4 Hz, H-1,10), 6.06 (d, $J = 12.5$ Hz, 1H, H-20), 5.97 (d, $J = 11.4$ Hz, 2H, H-2,9), 5.92 (d, $J =$ 12.5 Hz, 1H, H-18); ¹³C NMR δ 146.1 (C-16), 140.1 (C-8), 138.8 (C-22), 138.6 (C-3,7), 138.1 (C-21), 136.2 (C-11,15), 133.3 (C-19), 133.0 (C-1,10), 132.7 (C-2,9), 131.5 (C-20), 130.3 (C-5), 129.5 (C-18), 128.82 (C-13), 128.75 (C-17), 128.6 (25,26), 126.8 (C-12,14), 126.3 (C-4,6); IR ν (thin film) 3020, 1597, 1574, 1493, 1448, 960, 793, 763, 737, 691 cm⁻¹; UV−vis (cyclohexane) λ_{\max} nm (ε_{\max}) 252 (29300), 282 (26500), 408 (1600); EIMS m/z 408 (M⁺); HRMS calcd for $C_{32}H_{24}$ 408.1878, found 408.1884. On attempted melting point determination CPD 22 converted into DHP 23.

Eluted third was 67 mg (18%) of the trans−trans or bis-(1E) isomer 24: ¹H NMR δ 7.22–7.02 (m, 10H, H-23–32), 7.0 (t, J = 7.4 Hz, 2H, H-5,13), 6.86 (s, 4H, H-17−20), 6.62 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.48 (s, 4H, H-1,2,9,10); 13C NMR δ 144.6 (C-8,16), 138.2 (C-21,22), 136.7 (C-3,7,11,15), 133.6 (C-19,20), 133.3 (C-1,2,9,10), 129.3 (C-5,13), 128.8 (C-17,18), 127.4 (C-31,32), 127.0 (C-4,6,12,14), 126.8 (C-27,28,29,30), 126.5 (C-23,24,25,26); IR ν (KBr) 3026, 1598, 1492, 1447, 959, 838, 774, 763, 736, 690 cm[−]¹ ; UV−vis (cyclohexane) λmax nm (εmax) 258 (33000), 284 (33400), 308 (30300), 411 sh (1900). On attempted melting point determination CPD 24 converts into DHP 25.

10b,10c-Bis((1Z)-2-phenylvinyl)-trans-10b,10c-dihydropyrene (21). Thermolysis of cis,cis- styryl CPD 20, exactly as described for 12 above, gave 100% yield of the cis, cis-DHP 21 as orange crystals from cyclohexane: mp 152−154 °C; ¹H NMR (CD₂Cl₂) δ 8.26 (s, 4H, H-4,5,9,10), 8.21 (d, J = 7.5 Hz, 4H, H-1,3,6,8), 7.78 (t, J = 7.5 Hz, 2H, H-2,7), 7.15 (tt, J = 6.5, 1.1 Hz, 2H, H-25,26), 7.07−7.02 (m, 4H,

H-21−24), 6.03−6.01 (m, 4H, H-17−20), 4.21 (d, J = 12.6 Hz, 2H, 13,14), 0.28 (d, J = 12.6 Hz, 2H, H-11,12); ¹³C NMR (CD₂Cl₂) δ 136.8 (C-15,16), 134.2 (C-3a,5a,10a,10d), 128.8 (C-17−20), 127.5 (C-13,14), 126.9 (C-4,5,9,10), 126.4 (C-21−26), 125.7 (C-1,3,6,8), 124.1 (C-2,7), 122.5 (C-11,12), 36.7 (C-10b,10c); IR ν (KBr) 3024, 844, 830, 694, 609 cm⁻¹; UV-vis (cyclohexane) λ_{max} nm (ε_{max}) 351 nm (56400), 369 (15300), 395 (25700), 487 (5100), 624 (95); EIMS m/z 408 (M⁺); HRMS calcd for $C_{32}H_{24}$ 408.1878, found 408.1887. Anal. Calcd: C, 94.08; H, 5.92. Found: C, 93.90; H, 6.02.

10b-((1E)-2-Phenylvinyl)-10c-((1Z)-2-phenylvinyl)-trans-10b,10cdihydropyrene (23). Using the same procedure for cis,trans-styryl CPD 22 gave (∼100%) cis,trans-styryl DHP 23 as an oily film, which could not be crystallized: ¹H NMR δ 8.72 (d, J = 7.7 Hz, 2H, H-4,10), 8.64 (d, J = 7.7 Hz, 2H, H-1,3), 8.41 (d, J = 7.7 Hz, 2H, H-5,9), 8.27 (d, $J = 7.7$ Hz, 2H, H-6,8), 8.08 (t, $J = 7.6$ Hz, 1H, H-2), 7.81 (t, $J =$ 7.7 Hz, 1H, H-7), 7.18 (tt, $J = 7.4$, 1.0 Hz, 1H, H-25), 7.09 (dd, $J = 7.6$, 7.4 Hz, 2H, H-21,22), 6.78−6.68 (m, 3H, H-23,24,26), 6.11 (dd, J = 7.8, 0.9 Hz, 2H, H-17,18), 5.99 (dd, J = 7.8, 1.4 Hz, 2H, H-19,20), 4.31 $(d, J = 12.9 \text{ Hz}, 1H, H-13), 3.22 (d, J = 15.8 \text{ Hz}, 1H, H-14), 0.83 (d,$ $J = 15.9$ Hz, H-12), 0.41 (d, $J = 12.9$ Hz, 1H, H-11); ¹³C NMR δ 136.5 (C-15), 136.3 (C-16), 134.0 (C-3a,10a), 133.7 (C-5a,10d), 128.6 (C-17,18), 127.8 (C-23,24), 127.2 (C-13), 126.9 (C-5,9), 126.6 (C-26), 126.1 (C-21,22,25), 125.8 (C-6,8), 125.6 (C-19,20), 125.3 (C-4,10), 124.8 (C-14), 124.2 (C-1,3), 124.0 (C-7), 123.7 (C-2), 122.6 (C-11), 120.2 (C-12), 37.3 (C-10b), 15.0(C-10c); IR ν (thin film) 3028, 1733, 1597, 1490, 1441, 1355, 1070, 958, 836, 738, 693, 599 cm[−]¹ ; UV−vis (dichloromethane) λ_{max} nm (ε_{max}) 256 nm (20000), 348 (54800), 389 (25300), 478 (5780), 609 (106).

10b,10c-Bis((1E)-2-phenylvinyl)-trans-10b,10c-dihydropyrene (25). From the trans−trans-styryl CPD 24 using the same thermolysis procedure gave 100% of the trans,trans-styryl DHP 25 as dark green crystals from cyclohexane: mp 232-234 °C; ¹H NMR δ 8.85 (s, 4H, H-4,5,9,10), 8.67 (d, $J = 7.7$ Hz, 4H, H-1,3,6,8), 8.08 (t, $J = 7.7$ Hz, 2H, H-2,7), 6.76−6.69 (m, 6H, H-21−26), 6.03 (dd, J = 8.0, 2.5 Hz, 4H, H-17−20), 3.22 (d, J = 15.8 Hz, 2H, H-13,14), 0.87 (d, J = 15.8 Hz, 2H, H-11,12); ¹³C NMR δ 136.3 (C-15,16), 133.9 (C-3a,5a,10a,10d), 127.9 (C-21−24), 126.6 (C-25,26), 125.7 (C-17,20), 125.6 (C-4,5,9,10), 124.6 (C-1,3,6,8), 124.6 (C-13,14), 124.0 (C-2,7), 120.8 (C-11,12), 35.9 (C-10b,10c); IR ν (KBr) 3027, 1654, 1636, 1492, 1447, 959, 838, 774, 735, 717, 690 cm[−]¹ ; UV−vis (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 257 nm (47900), 342 (58000), 380 (26400), 466 (5770), 597 (165).

8,16-Bis(2-(4-nitrophenyl)ethenyl)-1,9-bis(methylthio)-anti-[2.2] metacyclophane (38). To a suspension of 4-nitrobenzyltriphenylphosphonium bromide (4.0 g, 8.4 mmol, commercial supplier) in toluene (50 mL) was added potassium tert-butoxide (1.3g, 12 mmol), the mixture was heated at 50 $^{\circ}$ C for 1 h, and then 8,16-diformyl-(methylthio)cyclophane 22 (1.0 g, 2.81 mmol) was added. The mixture was heated to reflux over a period of 6 h and stirred overnight. The toluene was then evaporated, and the crude was extracted with water and dichloromethane. The organic layer was dried and evaporated. The residue was chromatographed over silica gel using dichloromethane−hexane (55:45) as eluent and gave 0.90 g (54%) 38 as a yellow orange mixture of isomers: ¹H NMR δ 8.14 (d, J = 8.7 Hz), 7.99−7.96 (m), 7.94 (d, J = 8.6 Hz), 7.90−7.76 (m), 7.42−7.05 (m), 6.51−6.31 (m), 6.02−5.90 (m), 5.42−5.20 (m), 4.43−4.23 (m), 3.96− 3.85 (m), 3.17−3.05 (m), 2.98 (dd, J = 12.5, 3.9 Hz), 2.87 (t, J = 12.2 Hz), 2.68 (t, J = 12.5 Hz), 2.55−2.45 (m), 2.17, 2.11 (4s), 1.63 (s), 1.62 (s), 1.52 (s), 1.47 (s), 1.46 (s); HRMS calcd for $C_{34}H_{30}N_2O_4S_2$ 594.1647, found 594.1657. These mixed isomers were used directly in the next step.

8,16-Bis(2-(4-nitrophenyl)ethenyl)-1,9-bis(methylthio)-anti-[2.2] metacyclophane Bis-sulfonium Salt (42). Using the same procedure as for 33, the mixed isomers of 38 (900 mg, 1.50 mmol) on reaction with Borch reagent ((80% oil, 0.9 g, 4.4 mmol) in CH_2Cl_2 (22 mL) gave 1.20 g (quantitative) of bis-sulfonium salt 42: ¹H NMR (DMSO d_6) δ 8.22 (m), 7.89–7.39 (m), 6.61–6.19 (m), 4.60–4.54 (m), 4.32 $(d, J = 12.6 \text{ Hz})$, 4.18 $(d, J = 12.4 \text{ Hz})$, 3.44–3.32 (m), 3.26 (s), 3.25 (s), 3.23 (s), 3.20 (s), 2.95 (s), 2.94 (s), 2.92 (s), 2.87 (s), 2.81 (s). These were used in the next step.

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Bis-2-(4-nitrophenyl)vinyl)-anti-[2.2]metacyclophane-1,9-dienes (49−51). Using the same procedure as for 12, reaction of bissulfonium salt 42 (1.20 g, 1.5 mmol) and t-BuOK (600 mg, 5.4 mmol) in THF (10 mL) gave 350 mg (90%) of a mixture of cyclophanedienes 49−51. The residue was chromatographed over silica gel using hexanes/CH₂Cl₂ (60:40) as eluent. Eluted first was mostly the (ZZ) isomer 49 (\sim 75% pure). Washing the crystals with CH₂Cl₂ yielded 88 mg (23%) of pure (ZZ)-isomer as orange crystals: ^1H NMR δ 7.94 (d, $J = 8.8$ Hz, 4H, H-23,25,29,31), 7.23 (t, $J = 7.4$ Hz, 2H, H-5,13), 6.91 (d, $J = 8.8$ Hz, 4H, H-22, 26, 28, 32), 6.60 (d, $J = 7.4$ Hz, 4H, H-4,6,12,14), 6.00 (s, 4H, H-17−20), 5.84 (s, 4H, H-1,2,9,10); 13C NMR δ 146.2 (C-24,30), 145.8 (C-21,27), 140.8 (C-8,16), 138.2 (C-3,7,11,15), 132.6 (C-18,20), 132.5 (C-1,2,9,10), 130.9 (C-5,13), 129.4 (C-17,19), 129.2 (C-22,26,28,32), 126.6 (C-4,6,12,14), 123.3 (C-23,25,29,31); IR ν (KBr) 3045, 3008, 1592, 1512, 1339, 1106, 884, 855, 797, 772, 712 cm $^{-1}$; UV−vis (dichloromethane) $λ_{max}$ nm ($ε_{max}$) 254 (14900), 342 (16400), 423 sh (5800); EIMS m/z 498 (M⁺); HRMS calcd for $C_{32}H_{22}N_2O_4$ 498.1579, found 498.1593; attempted melting point determination converted CPD 49 into DHP 58.

Eluted second was 175 mg (45%) of the (ZE)-isomer 50 as orange red crystals from dichloromethane: ^1H NMR δ 8.13 (d, J = 8.8 Hz, 2H, H-29,31), 7.96 (d, J = 8.8 Hz, 2H, H-23,25), 7.30 (d, J = 8.8 Hz, 2H, H-28,32), 7.20 (t, J = 7.4 Hz, 1H, H-5), 6.98 (t, J = 7.4 Hz, 1H, H-13), 6.96 (d, J = 8.8 Hz, 2H, H-22,26), 6.95 (d, J = 16.2 Hz, 1H, H-19), 6.86 (d, J = 16.2 Hz, 1H, H-20), 6.76 (d, J = 7.4 Hz, 2H, H-4,6), 6.45 $(d, J = 7.4 \text{ Hz}, 2H, H-12, 14)$, 6.38 $(d, J = 11.4 \text{ Hz}, 2H, H-2, 9)$, 6.10 $(s,$ 2H, H-17,18), 6.00 (d, J = 11.4 Hz, 2H, H-1,10); ¹³C NMR δ 146.7 (C-30), 146.3 (C-24), 145.8 (C-21), 144.8 (C-8), 144.3 (C-27), 139.6 (C-16), 138.6 (C-11,15), 136.9 (C-3,7), 133.3 (C-2,9), 133.08 (C-1,10), 132.7 (C-18), 132.3 (C-19), 130.9 (C-13), 130.7 (C-20), 130.1 (C-5), 129.6 (C-17), 129.2 (C-22,26), 127.1 (C-4,6), 126.9 (C-28,32), 126.6 (C-12,14), 124.1 (C-29,31), 123.4 (C-23,25); IR ν (KBr) 3007, 1592, 1512, 1339, 1108, 855, 795, 772, 711 cm[−]¹ ; UV−vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 258 (32100), 340 (25000), 435 sh (9300). Attempted melting point determination converted CPD 50 into DHP 59.

Eluted third was 86 mg (22%) of the (EE)-isomer 51 as red crystals: ¹H NMR δ 8.17 (d, J = 8.8 Hz, 4H, H-23,25,29,31), 7.35 (d, J = 8.8 Hz, 4H, H-22,26,28,32), 7.05 (d, J = 16.2 Hz, 2H, H-17,19), 7.00 $(t, J = 7.4 \text{ Hz}, 2H, H - 5.13), 6.96 \text{ (d, } J = 16.2 \text{ Hz}, 2H, H - 18.20), 6.63$ $(d, J = 7.4 \text{ Hz}, 4\text{H}, \text{H-4}, 6, 12, 14)$, 6.55 (s, 4H, H-1,2,9,10); ¹³C NMR δ 146.8 (C-24,30), 144.4 (C-21,27), 143.7 (C-8,16), 137.0 (C-3,7,11,15), 133.8 (C-1,2,9,10), 132.5 (C-17,19), 131.1 (C-18,20), 130.3 (C-5,13), 127.3 (C-4,6,12,14), 127.0 (C-22,26,28,32), 124.1 (C-23,25,29,31); IR ν (KBr) 3046, 1588, 1507, 1335, 1108, 965, 863, 817, 761, 745, 688 cm⁻¹; UV−vis (dichloromethane) λ_{max} nm (ε_{max}) 240 (27000), 258 (25400), 285 sh (18100), 369 (28800), 444 sh (13400). Attempted melting point determination converted CPD 51 into DHP 60.

10b,10c-Bis((1Z)-2-(4-nitrophenyl)vinyl)-trans-10b,10c-dihydropyrene (58). Using the general thermolysis procedure above, (ZZ)-CPD 49 gave (ZZ)-DHP 58 quantitatively as orange crystals from dichloromethan: mp 243−244 °C; ¹H NMR δ 8.28 (s, 4H, H-4,5,9,10), 8.25 (d, $J = 7.5$ Hz, 4H, H-1,3,6,8), 7.94 (d, $J = 8.6$ Hz, 4H, H-21−24), 7.86 (t, J = 7.6 Hz, 2H, H-2,7), 6.20 (dd, J = 8.7, 0.8 Hz, 4H, H-17−20), 4.18 (d, J = 13.2 Hz, 2H, H-13,14), 0.38 (d, J = 13.2 Hz, 2H, H-11,12); ¹³C NMR δ 146.6 (C-25,26), 143.8 (C-15,16), 133.4 (C-3a,5a,10a,10d), 129.2 (C-17−20), 126.9 (C-4,5,9,10), 125.9 (C-1,3,6,8), 125.2 (C-13,14), 124.3 (C-2,7), 123.4 (C-11,12), 121.4 (C-21−24), 36.1 (C-10b,10c); IR ν (KBr) 3074, 3036, 1598, 1515, 1341, 1107, 854, 842 cm⁻¹; UV−vis (dichloromethane) $\lambda_{\rm max}$ nm $(\varepsilon_{\rm max})$ 281 (19500), 348 (46700), 397 (19500), 484 (5160); EIMS m/z 498 (M⁺); HRMS calcd for $C_{32}H_{22}N_2O_4$ 498.1580, found 498.1586.

10b-((1E)-2-(4-Nitrophenyl)vinyl)-10c-((1Z)-2-(4-nitrophenyl) vinyl)-trans-10b,10c-dihydropyrene (59). Using the general thermolysis procedure above, (ZE)-CPD 50 (25 mg) gave (ZE)-DHP 59 (25 mg, 100%) as olive green crystals from dichloromethane: mp 214−216 °C; ¹ H NMR δ 8.75 (d, J = 7.8 Hz, 2H, H-4,10), 8.68 (d, J = 7.7 Hz, 2H, H-1,3), 8.40 (d, J = 7.8 Hz, 2H, H-5,9), 8.29 (d, J = 7.7 Hz, 2H, H-6,8), 8.12 (t, J = 7.7 Hz, 1H, H-2), 7.97 (d, J = 8.6 Hz, 2H, H-21,22), 7.88 (t, J = 7.7 Hz, 1H, H-7), 7.56 (d, J = 9.0 Hz, 2H, H-23,24), 6.26 (d, J = 8.7 Hz, 2H, H-17,18), 6.08 (d, J = 9.0 Hz, 2H, H-19,20), 4.23 (d, $J = 13.1$ Hz, 1H, H-13), 3.25 (d, $J = 15.8$ Hz, 1H, H-14), 0.96 (d, J = 15.8 Hz, 1H, H-12), 0.46 (d, J = 13.1 Hz, 1H, H-11); 13C NMR δ 146.7 (C-25), 146.2 (C-26), 143.8 (C-15), 142.7 (C-16), 133.5 (C-3a,10a), 133.1 (C-5a,10d), 129.2 (C-17,18), 127.2 (C-5,9), 126.09 (C-19,20), 126.06 (C-6,8), 125.6 (C-4,10), 125.1 (C-12), 125.0 (C-13), 124.8 (C-1,3), 124.4 (C-2), 124.3 (C-7), 123.9 (C-11), 123.6 (C-14), 123.3 (C-23,24), 121.4 (C-21,22), 37.4 (C-10b), 34.8 (C-10c); IR v (KBr) 3031, 1595, 1512, 1341, 846 cm⁻¹; UV−vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 228 (32100), 344 (65800), 387 (25800), 481 (6860), 606 (125).

10b,10c-Bis((1E)-2-(4-nitrophenylvinyl)-trans-10b,10c-dihydropyrene (60). Using the general thermolysis procedure above, (EE)-CPD 51 (25 mg) gave (EE)-DHP 60 (25 mg, 100%) as dark green crystals from cyclohexane: mp 275−276 °C; ¹H NMR δ 8.88 (s, 4H, H-4,5,9,10), 8.71 (d, $J = 7.6$ Hz, 4H, H-1,3,6,8), 8.13 (t, $J = 7.6$ Hz, 2H, H-2,7), 7.60 (d, J = 8.8 Hz, 4H, H-21−24), 6.12 (d, J = 8.9 Hz, 4H, H-17−20), 3.28 (d, J = 15.7 Hz, 2H, H-13,14), 1.05 (d, J = 15.8 Hz, 2H, H-11,12); The compound was too insoluble to give a satisfactory 13C NMR spectrum. IR ν (KBr) 3032, 1513, 1340, 1109, 850 cm⁻¹; UV–vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 228 (16700), 340 (45900), 375 (16300), 473 (4800).

8,16-Bis((2-(4-methoxyphenyl)ethenyl)-1,9-bis(methylthio)-anti- [2.2]metacyclophane (37). Using the same procedure as for the synthesis of the p-nitro series (49) except that the reaction was heated to 80 °C overnight from 4-methoxybenzyltriphenylphosphonium bromide (3.89g, 8.4 mmol), potassium tert-butoxide (1.3g, 11.6 mmol), and diformylcyclophane (1.0g 2.81 mmol), in toluene (50 mL), there was obtained the cyclophanes 37 (795 mg, 1.4 mmol, 50%) as a mixture of isomers. Column chromatography over silica gel using dichloromethane−hexane 4:6 eluted the mixed isomers as a pale yellow solid: ¹ H NMR 7.95−7.80 (m), 7.29−7.20 (m), 7.04−6.98 (m) , 6.85−6.80 (m), 6.47−6.43 (m), 6.23 (d, J = 8.8 Hz), 6.18 (d, J = 9.0 Hz), 5.85−5.62 (m), 5.09−5.0 (m), 4.36 (dd), 4.09−3.93 (m), 3.801 (s), 3.800 (s), 3.798 (s), 3.66 (s), 3.65 (s), 3.63 (s), 3.62 (s), 3.12−3.00 (m), 2.91 (t), 2.72 (AB, J = 12 Hz), 2.59−2.50 (m), 2.17, 2.16, 2.10, 2.09, 1.64, 1.63, 1.52, 1.45, 1.44 (9 s)); EIMS m/z 564 (M⁺); HRMS calcd for $C_{36}H_{36}O_2S_2$ 564.2157, found 564.2169.

8,16-Bis((2-(4-methoxyphenyl)ethenyl)-1,9-bis(methylthio)-anti- [2.2]metacyclophane Bis-sulfonium Salt (41). Using the same procedure as for 33, mixed isomers 37 (200 mg, 0.35 mmol) on reaction with Borch reagent (80% oil, 0.5 g, 2.4 mmol) in CH_2Cl_2 (5 mL) gave 275 mg (quantitative) of bis-sulfonium salt 41: 1 H NMR $(DMSO-d₆)$ 7.84−7.34 (m), 7.25−7.17 (m), 6.99−6.91 (m), 6.63− 6.55 (m), 6.28−5.88 (m), 4.62−4.52 (m), 3.77, 3.62 (2 s), 3.48−3.20 (6 s), 2.97, 2.94, 2.92 (3 s), 2.36, 2.34, 2.24, 2.23, 1.99 (5 s), which were used directly in the next step.

Bis-2-(4-methoxyphenyl)vinyl)-anti-[2.2]metacyclophane-1,9-dienes (46−48). Using the same procedure as for 12, reaction of bissulfonium salt 41 (200 mg, 0.26 mmol) and t-BuOK (250 mg, 2.2 mmol) in THF (10 mL) gave a mixture of cyclophanedienes 46−48. The residue was chromatographed over silica gel using hexanes/DCM 60:40. Eluted first was 19 mg (16%) of the (ZZ) isomer 46 as colorless crystals: ¹H NMR δ 7.15 (t, J = 7.3 Hz, 2H, H-5,13), 6.74 (d, J = 8.7 Hz, 4H, H-22,26,28,32), 6.61 (d, J = 8.7 Hz, 4H, H-23,25,29,31), 6.57 (d, J = 7.3 Hz, 4H, H-4,6,12,14), 5.91 (d, J = 12.4 Hz, 2H, H-18,20), 5.86 (s, 4H, H-1,2,9,10), 5.73 (d, J = 12.4 Hz, 2H, H-17,19), 3.74 (s, 6H, H-33,34); 13C NMR δ 158.3 (C-24,30), 141.9 (C-8,16), 138.2 (C-3,7,11,15), 132.4 (C-1,2,9,10), 131.4 (C-21,27), 130.9 (C-18,20), 129.7 (C-22,26,28,32), 129.6 (C-5,13), 128.3 (C-17,19), 126.2 (C-4,6,12,14), 113.3 (C-23,25,29,31), 55.3 (C-33,34); IR ν (KBr) 3003, 2835, 1605, 1509, 1458, 1303, 1252, 1178, 1032, 832, 812, 784, 739, 517 cm⁻¹; UV-vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 228 (46400), 257 (46100), 372 sh (4160); EIMS m/z 468 (M⁺); HRMS calcd for $C_{34}H_{28}O_2$ 468.2089, found 468.2095. Attempted melting point determination converted CPD 46 into DHP 55.

Eluted second was 57 mg (47%) of the (ZE) isomer 47 as a pale yellow solid: ¹H NMR δ 7.17 (d, J = 8.8 Hz, 2H, H-22,26), 7.09 (t, J = 7.4 Hz, 1H, H-13), 7.02 (t, $J = 7.3$ Hz, 1H, H-5), 6.82 (d, $J = 8.8$ Hz, 2H, H-23,25), 6.78 (d, J = 8.7 Hz, 2H, H-28,32), 6.69 (d, J = 7.4 Hz,

2H, H-12,14), 6.68 (AB hidden, 2H, H-17,18), 6.65 (d, J = 8.6 Hz, 2H, H-29,31), 6.46 (d, J = 7.4 Hz, 2H, H-4,6), 6.32 (d, J = 11.3 Hz, 2H, H-1,10), 6.00 (d, J = 11.3 Hz, 2H, H-2,9), 5.98 (d, J = 12.3 Hz, 1H, H-20), 5.79 (d, J = 12.3 Hz, 1H, H-19), 3.82 (s, 3H, H-33), 3.75 (s, 3H, H-34); ¹³C NMR δ 159.2 (C-24), 158.3 (C-30), 146.5 (C-16), 140.3 (C-8), 138.6 (C-3,7), 135.9 (C-11,15), 133.1 (C-2,9), 132.8 (C-18), 132.5 (C-2,9), 131.3 (C-27), 131.1 (C-21), 131.0 (C-17), 130.1 (C-5), 129.8 (C-28,32), 128.28 (C-19), 128.24 (C-13), 128.0 (C-22,26), 126.7 (C-12,14), 126.3 (C-4,6), 114.0 (C-23,25), 113.3 (C-29,31), 55.5 (C-33), 55.3 (C-34); IR ν (KBr) 3031, 3001, 2834, 1604, 1509, 1465, 1250, 1175, 1033, 832, 812, 783 cm[−]¹ ; UV−vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 227 (44600), 257 (50000), 283 (36300), 417 sh (4900). Attempted melting point determination converted CPD 47 into DHP 56.

Eluted third was 27 mg (22%) of the (EE) isomer 48 as pale yellow crystals: ¹H NMR δ 7.20 (d, J = 8.8 Hz, 4H, H-22,26,28,32), 6.98 (t, $J = 7.4$ Hz, 2H, H-5,13), 6.84 (d, $J = 8.8$ Hz, 4H, H-23,25,29,31), 6.76 (AB, hidden, 4H, H-17−20), 6.60 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.45 (s, 4H, H-1,2,9,10), 3.83 (s, 6H, OMe); 13C NMR δ 159.2 (C-24,30), 144.8 (C-8,16), 136.4 (C-3,7,11,15), 133.14 (C-18,20), 133.11 (C-1,2,910), 131.2 (C-21,27), 128.9 (C-23,25,29,31), 128.3 (C-5,13), 128.0 (C-22,26,28,32), 127.0 (C-17,19), 126.9 (C-4,6,12,14), 55.5 (C-33,34); IR ν (KBr) 3031, 3001, 2833, 1604, 1509, 1465, 1438, 1250, 1174, 1034, 966, 810, 770, 753 cm⁻¹; UV−vis (dichloromethane) λ_{max} nm (ϵ_{max}) 228 (50600), 263 (45700), 417 sh (6200). Attempted melting point determination converted CPD 48 into DHP 57.

10b,10c-Bis((1Z)-2-(4-methoxyphenyl)vinyl)-trans-10b,10c-dihydropyrene (55). Using the general thermolysis procedure above, (ZZ)- CPD 48 (25 mg) gave (ZZ)-p-methoxystyryl DHP 55 (25 mg, 100%) as orange crystals from dichloromethane: ^1H NMR δ 8.28 (s, 4H, H-4,5,9,10), 8.24 (d, $J = 7.6$ Hz, 4H, H-1,3,6,8), 7.79 (t, $J = 7.6$ Hz, 2H, H-2,7), 6.59 (d, J = 8.6 Hz, 4H, H-21−24), 5.92 (d, J = 8.6 Hz, 4H, H-17−20), 4.12 (d, J = 12.8 Hz, 2H, H-13,14), 3.83 (OMe), 0.25 $(d, J = 12.8 \text{ Hz}, 2H, H-11, 12);$ ¹³C NMR (CD_2Cl_2) δ 158.6 (C-25,26), 134.3 (3a,5a,10a,10d), 129.8 (C-17−20), 129.1 (C-15,16), 127.3 (C-13,14), 126.8 (C-4,5,9,10), 125.7 (C-1,3,6,8), 124.0 (C-2,7), 122.9 (C-11,12), 111.8 (C-21,24), 55.7 (OMe), 36.8 (10b,10c); IR ν (thin film) 3030, 2834, 1605, 1574, 1507, 1463, 1243, 1174, 1034, 960, 860, 814, 737, 704 $\,\text{cm}^{-1}$; UVvis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 228 (34600), 351 (47100), 394 (25200), 486 (5500), 619 (140); EIMS m/z 468 (M⁺); HRMS calcd for $C_{34}H_{28}O_2$ 468.2089, found 468.2094.

10b-((1E)-2-(4-Methoxyphenyl)vinyl)-10c-((1Z)-2-(4-methoxyphenyl) vinyl)-trans-10b,10c-dihydropyrene (56). Using the general thermolysis procedure above, CPD 47 (25 mg), gave mostly (ZE)-p-methoxystyryl DHP 56 as an olive green solid containing a little of the (*EE*)-isomer (total 25 mg): ¹H NMR (CD₂Cl₂) δ 8.72 (d, J = 7.7 Hz, 2H, H-4,10), 8.64 (d, J = 7.7 Hz, 2H, 1,3), 8.44 (d, J = 7.7 Hz, 2H, H-5,9), 8.31 (d, $J = 7.7$ Hz, 2H, H-6,8), 8.06 (t, $J = 7.7$ Hz, 1H, H-2), 7.83 (t, J = 7.7 Hz, 1H, H-7), 6.62 (d, J = 8.5 Hz, 2H, H-21,22), 6.25 $(d, J = 8.8 \text{ Hz}, 2H, H-23,24)$, 5.98 $(d, J = 8.4 \text{ Hz}, 2H, H-17,18)$, 5.89 $(d, J = 8.8 \text{ Hz}, 2H, H-19,20), 4.20 (d, J = 12.8 \text{ Hz}, 1H, H-13), 3.84 (s,$ 3H, H-33), 3.48 (s, 3H, H-34), 3.15 (d, J = 15.8 Hz, 1H, H-14), 0.65 $(d, J = 15.8 \text{ Hz}, 1H, H-12), 0.33 (d, J = 12.8 \text{ Hz}, 1H, H-11);$ ¹³C NMR (CD_2Cl_2) δ 159.0 (C-26), 158.6 (C-25), 134.4 (C-3a,10a), 134.1 (C-5a,10d), 129.8 (C-17,18), 127.2 (C-5,9), 127.1 (C-13), 126.8 (C-19,20), 126.0 (C-6,8), 125.7 (C-4,10), 124.4 (C-1,3), 124.3 (C-14), 124.2 (C-7), 123.9 (C-2), 123.4 (C-11), 118.3 (C-12), 113.6 (C-23,24), 111.9 (C-21,22), 55.7 (C-33), 55.5 (C-34), 37.7 (C-10b), 35.6 (C-10c); IR ν (thin film) 3030, 2834, 1606, 1509, 1464, 1440, 1244, 1174, 1034, 959, 842, 815, 736, 704 cm[−]¹ ; UV−vis (dichloromethane) λ_{max} nm (ε_{max}) 225 (24300), 264 (32300), 346 (42100), 387 (21200), 472 (5400), 607 (120).

10b,10c-Bis((1E)-2-(4-methoxyphenylvinyl)-trans-10b,10c-dihydropyrene (57). Using the general thermolysis procedure above, CPD 48 (25 mg) gave p-methoxystyryl DHP 57 (25 mg, 100%) was obtained as a dark green oily solid, which would not crystallize: ¹H NMR (CD₂Cl₂) δ 8.88 (s, 4H, H-4,5,9,10), 8.70 (d, J = 7.7 Hz, 4H, 1,3,6,8), 8.10 (t, $J = 7.7$ Hz, 2H, H-2,7), 6.28 (d, $J = 8.8$ Hz, 4H, H-21−24), 5.95 (d, J = 8.8 Hz, 4H, H-17−20), 3.51 (OMe), 3.17 (d, $J = 15.8$ Hz, 2H, H-13,14), 0.70 (d, $J = 15.8$ Hz, 2H, H-11,12);

¹³C NMR (CD₂Cl₂) δ 159.0 (C-25,26), 134.5 (C-3a,5a,10a,10d), 129.8 (C-15,16), 126.8 (C-17−20), 126.0 (C-4,5,9,10), 124.8 $(C1,3,6,8)$, 124.21 $(C-13,14)$, 124.17 $(C-2,7)$, 119.0 $(C-11,12)$, 113.6 (C-21−24), 55.5 (OMe), 36.4 (C-10b,10c); IR ν (thin film) 3030, 2834, 1606, 1509, 1464, 1440, 1244, 1174, 1034, 959, 842, 815, 736, 704 cm⁻¹; UV−vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 225 (31000), 265 (40200), 343 (46800), 382 (22400), 470 (6400), 606 (130)

8,16-Bis(2-(4-methylphenyl)ethenyl)-1,9-bis(methylthio)-anti- [2.2]metacyclophane (36). Using the same procedure as for 12, reaction of diformylmethylthiocyclophane 22 (0.80 g, 2.25 mmol) with an ylide prepared by reaction of t-BuOK (1.7 g, 15 mmol) and 4 methylbenzyl triphenylphosphonium bromide (4.5 g, 10 mmol, commercial supplier) in THF (45 mL), followed by column chromatography using hexanes:dichloromethane (7:3) eluted 685 mg (57%) of 36 as a mixture of isomers: ¹H NMR δ 7.88 (d, J = 7.6 Hz), 7.84 (d, J = 7.7 Hz), 7.83−7.80 (m), 7.44−6.94 (m), 6.76−6.69 (m), 6.24 (d, $J = 8.2$ Hz), 6.20 (d, $J = 8.2$ Hz), 6.16 (d, $J = 8.6$ Hz), 6.14 (d, $J = 8.3$ Hz), 6.10 (d, $J = 8.2$ Hz), 5.92 (d, $J = 12.3$ Hz), 5.87 (d, $J = 12.4$ Hz), 5.82 (d, J = 12.4 Hz), 5.68 (d, J = 16.3 Hz), 5.23 (d, J = 16.3 Hz), 5.10 (d, J = 16.3 Hz), 4.35 (dd, J = 11.5, 4.0 Hz), 4.15−3.95 (m), 3.14− 2.87 (m), 2.75−2.51 (m), 2.36, 2.32, 2.28, 2.26 (4s), 2.17, 2.16, 2.15, 2.12, 2.11, 2.10, 2.03 (7s), 1.61, 1.60, 1.42, 1.40 (s)); EIMS m/z 532 (M⁺); HRMS calcd for $C_{36}H_{36}S_2$ 532.2258, found 532.2253. These were used directly in the next step.

8,16-Bis(2-(4-methylphenyl)ethenyl)-1,9-bis(methylthio)-anti- [2.2]metacyclophane Bis- sulfonium salt (40) . Using the same procedure as for 33, mixed isomers 36 (650 mg, 1.21 mmol) on reaction with Borch reagent (80% oil, 1.0 g, 4.8 mmol) in CH_2Cl_2 (7 mL) gave 510 mg (57%) of bis-sulfonium salt 40: ¹H NMR (DMSO- d_6) δ 7.83 (d, J = 7.7 Hz), 7.75 (d, J = 7.6 Hz), 7.71−7.50 (m), 7.47 (t, J = 7.6 Hz), 7.20−7.10 (m), 6.86−6.76 (m), 6.24−6.05 (m), 6.00−5.87 (m), 5.46 (d, J = 16.4 Hz), 5.30 (d, J = 16.6 Hz), 4.53 (br d), 4.26– 3.97 (m), 3.30, 3.22, 3.21, 3.20, 3.18 (s), 2.98−2.78 (m), 2.30, 2.28 (3 s), 2.18, 2.10, 1.96 (4 s). These were used in the next step.

Bis-2-(4-methylphenyl)vinyl)-anti-[2.2]metacyclophane-1,9-dienes (43−45). Using the same procedure as for 12, reaction of bissulfonium salt 40 (470 mg, 0.63 mmol) and t-BuOK (200 mg, 1.8 mmol) in THF (8 mL) gave a mixture of cyclophanedienes 43−45. The residue was chromatographed over silica gel using hexanes:dichloromethane (90:10). Eluted first was 116 mg (42%) (ZZ)-CPD 43 as colorless crystals: ¹H NMR δ 7.16 (t, J = 7.3 Hz, 2H, H-5,13), 6.88 (d, J = 8.0 Hz, 4H, H-23,25,29,31), 6.70 (d, J = 8.0 Hz, 4H, H-22,26,28,32), 6.57 (d, J = 7.3 Hz, 4H, H-4,6,12,14), 5.94 (d, J = 12.4 Hz, 2H, H-18,20), 5.85 (s, 4H, H-1,2,9,10), 5.78 (d, J = 12.4 Hz, 2H, H-17.19), 2.25 (s, 6H, Me); 13C NMR δ 141.8 (C-8,16), 138.2 (C-3,7,11,15), 136.0 (C-24,30). 135.8 (C-21,27), 132.4 (C-1,2,9,10), 131.3 (C-18,20), 129.6 (C-5,13), 128.9 (C-17,19), 128.6 (C-23,25,29,31), 128.4 (C-22,26,28,32), 126.1 (C-4,6,12,14), 21.4 (C-33,34); IR ν (KBr) 3041, 3015, 3003, 1560, 1508, 1370, 1405, 1151, 859, 817, 782, 749, 737, 594 cm⁻¹; UV−vis (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 257 (30100), 277 (24000), 392 (3300); EIMS m/z 436 (M⁺); HRMS calcd for $C_{34}H_{28}$ 436.2191, found 436.2205. Attempted melting point determination converted CPD 43 into DHP 52.

Eluted second was 105 mg (38%) of the (ZE)-isomer 44 as pale yellow crystals from cyclohexane: ^1H NMR δ 7.12 (d, J = 8.3 Hz, 2H, H-22,26), 7.10 (t, J = 7.5 Hz, 1H, H-13), 7.08 (d, J = 8.4 Hz, 2H, H-23,25), 7.01 (t, $J = 7.4$ Hz, 1H, H-5), 6.90 (d, $J = 8.0$ Hz, 2H, H-29,31), 6.86−6.73 (m, 4H, H-17,18,28,32), 6.71 (d, J = 7.4 Hz, 2H, H-12,14), 6.47 (d, $J = 7.4$ Hz, 2H, H-4,6), 6.33 (d, $J = 11.3$ Hz, 2H, H-1,10), 6.02 (d, J = 12.3 Hz, H-20), 6.00 (d, J = 11.4 Hz, 2H, H-2,9), 5.83 (d, J = 12.3 Hz, 1H, H-19). 2.34 (s, 3H, H-33), 2.26 (s, 3H, H-34); 13C NMR δ 146.3 (C-16), 140.2 (C-8), 137.2 (C-24), 138.6 (C-3,7), 136.09 (C-30), 136.07 (C-11,15), 135.8 (C-27), 133.2 (C-18), 133.0 (C-1,10), 132.6 (C-2,9), 131.5 (C-20), 130.2 (C-5), 129.2 (C-13), 128.9 (C-19), 128.6 (C-29,31), 128.5 (C-28,32), 127.8 $(C-17)$, 126.8 $(C-12,14)$, 126.7 $(C-22,26)$, 126.3 $(C-4,6)$, 21.5 $(C-33)$, 21.4 (C-34) IR ν (thin film) 3045, 3019, 3005, 1510, 1436, 1265, 963, 868, 816, 805, 786, 777, 746, 702, 647 cm⁻¹; UV-vis (cyclohexane)

 λ_{max} nm $(\varepsilon_{\text{max}})$ 252 (60000), 280 (48400), 413 sh (6600). Attempted melting point determination converted CPD 44 into DHP 53.

Eluted third was 30 mg (11%) of the (EE)-isomer 45 as pale yellow crystals: ¹ H NMR δ 7.16 (AB, J = 8.1 Hz, 4H, H-22,26,28,32), 7.10 $(AB, J = 8.1 \text{ Hz}, 4H, H-23,25,29,31), 6.98 \text{ (t, } J = 7.3 \text{ Hz}, 2H, H-5,13),$ 6.86 and 6.84 (AB, J = 16.2, 4H, H-17−20), 6.60 (d, J = 7.3 Hz, 4H, H-4,6,12,14), 6.46 (s, 4H, H-1,2,9,10) 2.34 (s, 6H, Me); ¹³C NMR δ 144.7 (C-8,16), 137.2 (C-24,30), 136.5 (C-3,7,11,15), 135.5 (C-21,27), 133.5 (C-18,20), 133.2 (C-1,2,9,10), 129.2 (C-23,25,29,31), 129.1 (C-5,13), 127.9 (C-17,19), 126.9 (C-4,6,12,14), 126.7 (C-22,26,28,32). 21.5 (C-33,34, Me) ; IR ν (KBr) 3045, 3022, 3003, 1560, 1513, 1458, 1429, 960, 797, 755, 653, 631 cm[−]¹ ; UV−vis (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 257 (53500), 285 (53000), 313 (48700), 410 sh (6100). Attempted melting point determination converted CPD 45 into DHP 54.

10b,10c-Bis((1Z)-2-(4-methylphenyl)vinyl)-trans-10b,10c-dihydropyrene (52). Using the general thermolysis procedure above, (ZZ)-CPD 43 (25 mg) gave (ZZ) -p-methylstyryl DHP 52 (25 mg, 100%) as orange crystals from cyclohexane: ${}^{1}H$ NMR (CD₂Cl₂) δ 8.25 (s, 4H, H-4,5,9,10), 8.22 (d, $J = 7.5$ Hz, 4H, H-1,3,6,8), 7.78 (t, $J = 7.5$ Hz, 2H, H-2,7), 6.84 (d, J = 7.6 Hz, 4H, H-21−24), 5.89 (d, J = 7.5 Hz, 4H, H-17−20), 4.17 (d, J = 12.9 Hz, 2H, H-13,14), 2.34 (s, 6H, Me) 0.25 (d, J = 12.9 Hz, 2H, H-11,12); ¹³C NMR (CD₂Cl₂) δ 135.8 (C-25,26), 134.2 (C-3a,5a,10a,10d), 133.7 (C-15,16), 128.6 (17−20), 127.5 (C-13,14), 127.0 (C-21−24), 126.9 (C-4,5,9,10), 125.7 (C-1,3,6,8), 124.0 (C-2,7), 122.5 (C-11,12), 36.7 (C-10b,10c), 21.5 (Me); IR ν (KBr) 3038, 3019, 1507, 1345, 860, 836, 607 cm[−]¹ ; UV−vis (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 227 (37500), 353 (68800), 396 (33300), 490 (6300), 624 (130); EIMS m/z 436 (M⁺); HRMS calcd for $\rm C_{34}H_{28}$ 436.2191, found 436.2197.

10b-((1E)-2-(4-Methylphenyl)vinyl)-10c-((1Z)-2-(4-methylphenyl) vinyl)-trans-10b,10c-dihydropyrene (53). Using the general thermolysis procedure above, (ZE)-CPD 44 (25 mg) gave (ZE)-p-methylstyryl DHP 53 (25 mg, 100%) as an olive green oily film which could not be crystallized: ¹H NMR (CD₂Cl₂) δ 8.72 (d, J = 7.7 Hz, 2H, H-4,10), 8.63 (d, J = 7.7 Hz, 2H, H-1,3), 8.41 (d, J = 7.7 Hz, 2H, H-5,9), 8.28 (d, J = 7.7 Hz, 2H, H-6,8), 8.06 (t, J = 7.7 Hz, 1H, H-2), 7.82 (t, J = 7.7 Hz, 1H, H-7), 6.89 (d, J = 7.7 Hz, 2H, H-21,22), 6.51 (d, J = 8.1, 2H, H-23,24), 5.95 (d, J = 7.4 Hz, 2H, H-17,18), 5.84 (d, J = 8.3 Hz, 2H, H-19,20), 4.22 (d, $J = 12.8$ Hz, 1H, H-13), 3.17 (d, $J = 15.9$ Hz, 1H, H-14), 2.36 (s, 3H, H-33), 1.96 (s, 3H, H-34), 0.73 (d, J = 15.9 Hz, 1H, H-12), 0.32 (d, J = 12.9 Hz, 1H, H-11); 13C NMR (CD_2Cl_2) δ 136.9 (C-26), 135.9 (C-25), 134.4 (C-3a,10a), 134.1 (C-5a,10d), 133.7 (C-15), 133.6 (C-16), 128.9 (C-23,24), 128.7 (C-17,18), 127.4 (C-13), 127.3 (C-5,9), 127.1 (C-21,22), 126.1 (C-6,8), 125.5 (C-19,20), 124.9 (C-14), 124.4 (C-1,3), 124.3 (C-7), 123.9 (C-2), 122.9 (C-11), 119.6 (C-12), 37.7 (C-10b), 35.5 (C-10c), 21.5 (C-33), 21.0 (C-34); IR ν (thin film) 3025, 2921, 2851, 1637, 1512, 1458, 1123, 962, 840, 801, 711 cm[−]¹ ; UV−vis (dichloromethane) λ_{\max} nm (ϵ_{\max}) 227 (24500), 257 (28900), 347 (42900), 387 (19100), 478 (4300), 607 (100).

10b,10c-Bis((1E)-2-(4-methylphenylvinyl)-trans-10b,10c-dihydropyrene (54). Using the general thermolysis procedure above, (EE)-CPD 45 (25 mg) gave (EE)-p-methylstyryl DHP 54 (25 mg, 100%) as a dark green oily solid which could not be crystallized: ¹H NMR (CD_2Cl_2) δ 8.88 (s, 4H, H-4,5,9,10), 8.69 (d, J = 7.6 Hz, 4H, H-1,3,6,8), 8.10 (t, $J = 7.6$ Hz, 2H, H-2,7), 6.55 (d, $J = 8.2$ Hz, 4H, H-21−24), 5.89 (d, J = 8.2 Hz, 4H, H-17−20), 3.19 (d, J = 15.8 Hz, 2H, H-13,14), 1.99 (s, Me), 0.79 (d, J = 15.8 Hz, 2H, H-11,12); ¹³C NMR (CD_2Cl_2) δ 137.0 $(C-25,26)$, 134.4 $(C-3a,5a,10a,10d)$, 133.5 (C-15,16), 128.9 (C-21−24), 126.0 (C-4,5,9,10), 125.6 (C-17− 20), 124.9 (C-1,3,6,8), 124.7 (C-13,14), 124.3 (C-2,7), 120.1 (C-11,12), 36.3 (C-10b,10c) 21.0 (Me); IR ν (KBr) 3025, 2921, 2851, 1637, 1512, 1458, 962, 840, 801, 711 cm[−]¹ ; UV−vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 227 (19100), 259 (33700), 343 (40300), 380 (21000), 468 (5400), 593 (170).

8,16-Bis(4-methylpenta-1,3-dienyl)-1,9-bis(methylthio)-anti- [2.2]metacyclophane (92).

Using the same procedure as for 12, reaction of diformylmethylthiocyclophane 22 (680 mg, 1.91 mmol) with an ylide prepared by reaction of t-BuOK (1.60 g, 14.3 mmol) and 3-methylbut-2-enyltriphenylphosphonium bromide (4.2 g, 10.2 mmol) in THF (40 mL), followed by column chromatography using hexanes/dichloromethane (75:25) gave 590 mg (67%) of 92 as a mixture of isomers: ¹H NMR δ 7.88−7.78 (m), 7.24−6.89 (m), 5.86−5.56 (m), 5.50−5.40 (m), 5.15− 4.91 (m), 4.70−4.56 (m), 4.40−4.32 (m), 4.10−3.95 (m), 3.10−3.00 (m), 2.91−2.83 (m), 2.70−2.50 (m), 2.28 (t, J = 11.4 Hz), 2.19, 2.18 (s), 2.13, 2.09 (3 s), 2.06, 2.04, 2.03 (3 s), 1.74, 1.61, 1.60 (br s), 1.53, 1.51 (br s)); EIMS m/z 460 (M⁺); HRMS calcd for C₃₀H₃₆S₂ 460.2258, found 460.2257. These were used directly in the next step 8,16-Bis(4-methylpenta-1,3-dienyl)-1,9-bis(methylthio)-anti-

[2.2]metacyclophane Bis-sulfonium Salt (93).

Using the same procedure as for 33, mixed isomers of 92 (0.55 g, 1.2 mmol) on reaction with Borch reagent ((80% oil, 0.9 g, 4.4 mmol) in CH_2Cl_2 (8 mL) gave 320 mg (41%) of bis-sulfonium salt 93: ¹H NMR $(DMSO-d₆)$ 7.77–7.32 (m), 7.26 (t, J = 7.6 Hz), 7.10 (t, J = 7.5 Hz), 6.08−5.92 (m), 5.82−5.71(m), 5.06 (t, $I = 12.2$ Hz), 4.93−4.74 (m), 4.63−4.42 (m), 4.18 (d, J = 11.2 Hz), 4.12 (d, J = 11.7 Hz) 4.04 and 4.02 (AB, J = 14.2, 7.0 Hz), 3.9 (d, J = 11.2 Hz), 3.52–3.10 (m), 3.32– 3.29 (overlapping singlets), 2.91−2.76 (6 s), 1.75, 1.64, 1.62, 1.57, 1.55, 1.52 (br s). These were used in the next step.

8,16-Bis(4-methylpenta-1,3-dienyl)-anti-[2.2]metacyclophane-1,9-dienes (62−64). Using the same procedure as for 12, reaction of bis-sulfonium salt 93 (300 mg, 1.05 mmol) and t-BuOK (400 mg, 3.6 mmol) in THF (20 mL) gave a mixture of cyclophanedienes (325 mg, 85%, approximately 1:1:1 isomer mixture), which was chromatographed over silica gel using hexanes/dichloromethane (19:1). Eluted first from column was a mixture of (EE)-isomer 64 and (ZE)-63 and then last 63 and the (ZZ)-isomer 62. Although the isomers could not be obtained completely free of each other, their spectral properties could be obtained:

(EE)-Isomer 64: ¹H NMR (CD₂Cl₂) δ 6.98 (t, J = 7.4 Hz, 2H, H-5,13), 6.80 (dd, J = 15.3, 11.0 Hz, 2H, H-19,20), 6.53 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.38 (s, 4H, H-1,2,9,10), 6.09 (d, J = 15.4 Hz, 2H, H-17,18), 5.60 (dm, J = 10.8 Hz, 2H, H-21,22), 1.78 (s, 6H, H-25,28), 1.72 (s, 6H, H-26,27); ¹³C NMR (CD₂Cl₂) δ 146.0 (C-8,16), 136.4 (C-3,7,11,15), 136.3 (C-23,24), 133.5 (C-1,2,9,10), 130.9 (C-19,20), 129.8 (C-17,18), 129.0 (C-5,13), 127.1 (C-21,22), 126.9 (C-4,6,12,14), 26.5 (C-25,28), 18.7 (C-26,27); EIMS m/z 364 (M⁺); HRMS calcd for C28H28 364.2191, found 364.2195.

(EZ)-Isomer **63**: ¹H NMR (CD₂Cl₂) δ 7.09 (t, J = 7.4 Hz, 1H, H-13), 6.99 (t, $J = 7.4$ Hz, 1H, H-5), 6.75 (dd, $J = 15.2$, 11.0 Hz, 1H, H-20), 6.58 (d, J = 7.4 Hz, 2H, H-12,14), 6.57 (d, J = 7.4 Hz, 2H, H-4,6), 6.33 and 6.31 (AB, $J = 11.4$ Hz, 4H, H-1,2,9,10), 6.04 (d, $J = 15.4$ Hz, 1H, H-18), 6.00−5.95 (m, 1H, H-21), 5.83 (dd, J = 12.4, 11.8 Hz, 1H, H-19), 5.79 (d, J = 12.4 Hz, 1H, H-17), 5.60 (dm, J = 12.5 Hz, 1H, H-22), 1.78 (s, 3H, H-28), 1.74 (s, 3H, H-25), 1.72 (s, 3H, H-27), 1.71 (s, 3H, H-26); ¹³C NMR (CD₂Cl₂) δ 147.1 (C-8), 142.4 (C-16), 138.2 (C-11,15), 137.4 (C-23), 136.5 (C-24), 136.2 (C-3,7), 133.1 (C-2,9), 133.0 (C-1,10), 130.9 (C-20), 129.8 (C-13), 129.6 (C-18),

128.4 (C-5), 127.1 (C-22), 126.89 (C-4,6), 126.87 (C-17), 126.42 (C-19), 126.40 (C-12,14), 123.4 (C-21), 26.6 (C-25), 26.5 (C-28), 18.7 (C-27), 18.6 (C-26).

(ZZ)-Isomer **62.** ¹H NMR (CD₂Cl₂) δ 7.10 (t, J = 7.4 Hz, 2H, H-5,13), 6.63 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.24 (s, 4H, H-1,2,9,10), 6.00−5.97 (m, 2H, H-21,22), 5.76 (br s, 4H, H-17−20), 1.74 (s, 6H, H-25,28), 1.71 (s, 6H, H-26,27); ¹³C NMR (CD₂Cl₂) δ 143.4 (C-8,16), 137.9 (C-3,7,11,15), 137.6 (C-23,24), 132.6 (C-1,2,9,10), 129.2 (C-5,13), 126.6, 126.5 (C-17−20), 126.3 (C-4,6,12,14), 26.6 (C-25,28), 18.6 (C-26,27); IR (mix) ν (KBr) 3041, 3002, 2965, 2910, 1636, 1560, 426, 1375, 952, 786, 777, 750 cm⁻¹; UV−vis (cyclohexane) λmax (εmax) 246 (27900), 284 (32600), 390 (5000), 406 sh (4800).

10b,10c-Bis(4-methylpenta-1,3-dienyl)-trans-10b,10c-dihydropyrene Dihydropyrenes (65−67). A mixture of cyclophanedienes 62−64 (45 mg) in toluene- d_8 (6 mL) was sealed in a glass tube under argon and heated at 110 °C for 2 h, which converted it into a mixture of dihydropyrenes 65−67 (45 mg, approximately 1:1:1). The mixture was chromatographed over silica gel using hexanes-dichloromethane $(9:1)$ as eluent. Eluted first was the (EE) -isomer 67, dark green colored crystals: mp 188−190 °C; ¹H NMR δ 8.76 (s, 4H, H-4,5,9,10), 8.63 (d, J = 7.8 Hz, 4H, H-1,3,6,8), 8.06 (t, J = 7.7 Hz, 2H, H-2,7), 4.04 (dm, $J = 10.9$ Hz, 2H, H-15,16), 2.95 (dd, $J = 15.0$, 10.8 Hz, 2H, H-13,14), 1.15, 1.10 (s, 12H, Me), 0.22 (d, J = 15.0 Hz, 2H, H-11,12); 13C NMR δ 134.4 (C-3a,5a,10a,10d), 133.5 (C-17,18), 125.3 (C-4,5,9,10), 124.2 (C-1,3,6,8), 123.8 (C-15,16), 123.7 (C-2,7), 121.8 (C-11,12), 121.4 (C-13,14), 35.8 (C-10b,10c), 25.5, 18.0 (C19− 22); IR ν (KBr) 3053, 3029, 3003, 1654, 1648, 1441, 1376, 956, 843, 710 cm[−]¹ ; UV−vis (cyclohexane) λmax nm (εmax) 247 (76200), 338 (78300), 380 (27900), 469 (7900), 609 (135) EIMS m/z 364 (M⁺); HRMS calcd for $C_{28}H_{28}$ 364.2191, found 364.2195.

Eluted second was mostly the (\textit{ZE}) -isomer 66: ^1H NMR δ 8.72 and 8.69 (AB, J = 7.7 Hz, 4H, H-4,5,9,10), 8.69 (d, J = 7.6 Hz, 2H, H-6,8), 8.60 (d, J = 7.8 Hz, 2H, H-1,3), 8.08 (t, J = 7.7 Hz, 1H, H-7), 8.04 (t, $J = 7.8$ Hz, 1H, H-2), 4.50 (dm, $J = 12$, 1.3 Hz, 1H, H-15), 4.06 (dm, $J = 10.8$ Hz 1H, H-16), 3.84 (t, $J = 12.3$ Hz, 1H, H-13), 3.03 (dd, $J =$ 15.0, 10.8 Hz, 1H, H-14), 1.52, 1.00 (br s, 6H, H-19,20), 1.16. 1.11 $(2s, 6H, H-21, 22), 0.23$ $(d, J = 15.0$ Hz, 1H, H-12), 0.01 $(d, J = 12.4)$ Hz, 1H, H-11); ¹³C NMR 135.9 (C-17,18), 135.3 (C-5a,10d), 134.8 (3a,10a), 126.3 (C-5,9), 125.4 (C-4,10), 125.1 (C-6,8), 124.4 (C-1,3), 123.84 (C-16), 123.78 (C-2), 123.2 (C-7), 122.7 (C-13), 121.9 (C-14), 120.8 (C-12), 120.0 (C-11), 119.4 (C-15), 37.8 (C-10b), 34.8 (C-10c), 26.5, 17.1 (C-19,20), 25.5, 18.0 (C-21,22); IR ν (KBr) 3030, 3003, 2926, 2853, 1654, 1438, 1374, 955, 843, 838, 714, 645 cm⁻¹; UV−vis (cyclohexane) $λ_{max}$ nm ($ε_{max}$) 250 nm (54800), 344 (56400), 389 (30400), 475 (6058), 614 (120).

Eluted third was mostly the (ZZ)-isomer 65: ¹H NMR δ 8.65 (d, $J = 7.7$ Hz, 4H, H-1,3,6,8), 8.62 (s, 4H, H-4,5,9,10), 8.08 (2H, H-2,7), 4.53 (dm, J = 12.2, 1.4 Hz, 2H, H-15,16), 3.90 (dd, J = 12.2 Hz, 2H, H-13,14), 1.52, 1.01 (s, 12H, Me), −0.02 (d, J = 12.5 Hz, 2H, H-11,12); 13C NMR δ 136.1 (C-17,18), 135.9 (C-3a,5a,10a,10d), 126.3 (C-4,5,9,10), 125.4 (C-1,3,6,8), 123.5 (C-2,7), 123.3 (C-13,14), 119.4 (C-15,16), 119.0 (C-11,12), 36.6 (C-10b,10c), 26.6, 17.1 (C-19−22); IR ν (KBr) 3023, 2965, 1647, 1654, 1438, 1374, 836, 729 cm^{−1}; UV (cyclohexane) λ_{max} nm (ε_{max}) 251 (42900), 347 (38200), 392 (20100), 486 (5000), 627 (137).

8,16-Diethynyl-anti-[2.2]metacyclophane-1,9-diene (69). Using the procedure above for 32, from the ylide prepared by reaction of t-BuOK (3.0 g, 27 mmol) and bromomethyl triphenylphosphonium bromide (2.76 g, 6.3 mmol) in THF (25 mL) and diformylmethylthiocyclophane 22 (1.0 g, 2.81 mmol) and column chromatography using hexanes: dichloromethane (7:3) there was obtained 600 mg (see below for yield) of a mixture of isomers containing 68: ^1H NMR δ 7.86 (d, J = 7.6 Hz), 7.84 (d, J = 7.7 Hz), 7.67−7.62 (m), 7.55−7.52 (m), 7.47−7.41 (m), 7.30−7.25 (m), 7.19 (t, J = 7.5 Hz), 7.06 (t, J = 7.3 Hz), 5.83−5.73 (m), 4.58 (d, J = 7.8 Hz), 4.54 (d, J = 7.6 Hz), 3.89 (dd, J = 11.5, 4.0 Hz), 3.80 (dd, J = 711.5, 3.9 Hz), 3.17–3.13 (m), 2.55 (t, J = 12.5 Hz), 2.44 (t, J = 12.1 Hz), 2.17, 2.15, 2.12, 2.10 (4 s); ¹³C NMR δ 144.4, 143.0, 141.7, 141.2, 136.1, 135.90, 135.87, 134.9, 133.2, 132.7, 132.3, 132.14, 132.12, 132.0, 131.4, 130.2, 129.8, 129.6,

129.2, 128.9, 128.8, 128.7, 128.5, 128.4, 128.0, 127.9, 127.6, 127.5, 127.3, 127.1, 126.3, 125.9, 125.2, 124.9, 123.5, 110.2, 109.5, 109.3, 109.04, 108.96, 87.2, 54.0, 53.5, 53.3, 53.0, 50.7, 46.2, 44.12, 44.05, 43.6, 29.9, 28.4, 22.9, 21.1, 17.6, 17.0, 16.9, 15.9, 15.6, 14.3. This material was used directly in the next step using the same procedure used to prepare 33. From these mixed isomers (600 mg) on reaction with Borch reagent (80% oil, 1.0 g, 4.84 mmol) in CH_2Cl_2 (7 mL) there was obtained 600 mg of the bis-sulfonium salt of $68:$ ¹H NMR $(DMSO-d_6)$ δ 7.81 (d, J = 7.8 Hz), 7.69 (d, J = 7.5 Hz), 7.62 (d, J = 7.6 Hz), 7.52 (d, J = 7.6 Hz), 7.41 (t, J = 7.6 Hz), 7.27 (t, J = 7.2 Hz), 6.41 (d, J = 7.6 Hz), 6.36 (d, J = 7.8 Hz), 6.25 (d, J = 8.0 Hz), 5.08 (d, J = 7.7 Hz), 4.74 (d, J = 7.8 Hz), 4.57 (dd, J = 11.2, 3.3 Hz), 4.48−4.42 (m), 3.55−3.50 (m), 3.33, 3.32 (2s), 2.86, 2.84 (s), 2.93−2.74 (m). These were then suspended in THF (20 mL) and t-BuOK (600 mg, 5.35 mmol) was added and the mixture stirred at 20 °C for 2h. Water and $CH₂Cl₂$ were added and the organic extract was washed, dried and evaporated. Chromatography over silica gel using hexane−CH2Cl2 90:10 gave the colorless diethynyl CPD 69 (100 mg, ~14% overall). However, the yield was variable from run to run: ^{1}H NMR δ 7.19 (t, $J = 7.5$ Hz, 2H, H-5,13), 6.75 (d, $J = 7.5$ Hz, 4H, H-4,6,12,14), 6.49 $(s, 4H, H-1, 2, 9, 10), 2.84 (H-18, 20);$ ¹³C NMR δ 141.1 (C-3,7,11,15), 133.9 (C-1,2,9,10), 131.4 (C-5,13), 129.8 (C-8,16), 126.5 (C-4,6,12,14), 834 (C-18,20), 81.8 (C-17,19); IR ν (KBr) 3293, 3050, 3005, 1727, 1427, 1262, 857, 805, 755, 655 cm⁻¹; UV-vis (dichloromethane) λ_{\max} nm $(\varepsilon_{\text{max}})$ 235 (35800), 281 (12900); EIMS m/z 252 (M⁺, 100%); HRMS calcd for $C_{20}H_{12}$ 252.0939, found 252.0923. On attempted melting point determination, diethynyl CPD 69 isomerized to diethynyl DHP 70.

10b,10c-Diethynyl-trans-10b,10c-dihydropyrene (70). Using the general thermolysis procedure above, diethynyl CPD 69 (25 mg) gave diethynyl DHP 70 (25 mg, 100%) as dark green crystals from CH_2Cl_2 : mp 245−250 °C dec; ¹H NMR δ 8.88 (s, 4H, H-4,5,9,10), 8.74 (d, J = 7.6 Hz, 4H, H-1,3,6,8), 8.27 (t, J = 7.7 Hz, 2H, H-2,7), −0.09 (s, 2H, H-12,14); 13C NMR δ 132.9 (C-3a,5a,10a,10d), 125.5 (C-4,5,9,10), 124.6 (C-2,7), 123.8 (C-1,3,6,8), 75.4 (C-11,13), 65.0 (C-12,14), 30.0 $(C-10b, 10c)$; IR ν (KBr) 3264, 3042, 1654, 1353, 845, 756, 662 cm⁻¹; UV–vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 223 (2900), 229 (7900), 336 (9700), 372 (42700), 399 (4400), 452 (7200), 535 (100), 596 (130), 639 (210); EIMS m/z 252 (M⁺); HRMS calcd for $C_{20}H_{12}$ 252.0939, found 252.0923.

6-tert-Butyl-18-cyano-9-methyl-2,11-dithia[3.3]metacyclophanes (73) and (74). A solution of 2,6-bis(bromomethyl)benzonitrile¹¹ 79 (4.40 g, 15.2 mmol) and 2,6 bis(mercaptomethyl)-4-tert-butylto-luene¹² 72 (3.80 g, 15.2 mmol[\)](#page-14-0) in deaerated benzene (700 mL) was added dropwise under nitrogen to an ethanolic KOH solution, prep[are](#page-14-0)d by adding KOH (4.83 g) to deaerated water (230 mL) and ethanol (2090 mL) followed by addition of sodium borohydride (0.9 g). The drop rate was crucial (∼1 every 3−4 s), and the addition took 44 h. The solvent was then evaporated and the residue was dissolved in dichloromethane and was washed with water, dried, and evaporated. The residue was chromatographed on silica gel using CH_2Cl_2 −hexanes $(35:65)$ as eluent. Eluted first was 1.90 g $(34%)$ of *anti*-thiacyclophane 74 as colorless crystals from cyclohexane: mp 147-148 °C; ¹H NMR δ 7.42−7.45 (m, 3H, H-14,15,16), 7.43 (s, 2H, H-5,7), 3.92 and 3.72 $(AB, J = 14.4 \text{ Hz}, 4H, H-1, 12), 3.82 \text{ and } 3.74 (AB, J = 14.0 \text{ Hz}, 4H, H-1, 12)$ 3,10), 1.43 (s, 3H, Me), 1.39 (s, 9H, C(CH₃)₃); ¹³C NMR δ 149.3 (C-6), 141.5 (C-13,17), 136.6 (C-9), 133.5 (C-4,8), 131.9 (C-15), 129.6 $(C-14,16)$, 128.2 $(C-5,7)$, 115.4 (CN) , 115.1 $(C-18)$, 34.6 $(C(CH_3)$ ₃), 33.0 (C-3,10), 31.6 (C-1,12), 31.3 (C(CH₃)₃), 15.4 (C-20); IR ν (KBr) 3027, 2954, 2217, 1588, 1483, 1464, 1417, 1236, 877, 804, 794, 755 cm[−]¹ ; UV (cyclohexane) λmax nm (εmax) 297 (1410), 307 (1390); EIMS m/z 367 (100); HRMS calcd for $C_{22}H_{25}NS_2$ 367.1428, found 367.1413. Anal. Calcd: C, 71.89; H, 6.86; N, 3.81. Found: C, 71.52; H, 6.87; N, 3.69.

Eluted second using CH_2Cl_2 −hexanes (45:55) was 1.80 g (32%) of the syn-thiacyclophane 73 as colorless crystals from cyclohexane: mp 166−167 °C; ¹ H NMR δ 7.07 (br s, 3H, H-14−16), 6.96 (s, 2H, H-5,7), 4.36 and 3.75 (AB, J = 15.1 Hz, 4H, H-1,12), 4.36 and 3.67(AB, J = 15 Hz, 4H, H-3,10), 2.56 (s, 3H, Me), 1.18 (s, 9H, $C(CH_3)_3$; ¹³C NMR δ 147.4 (C-6), 142.5 (C-13,17), 135.9 (C-4,8),

132.9 (C-15), 132.2 (C-9), 128.7 (C-14,16), 126.3 (C-5,7), 117.7 (CN), 111.7 (C-18), 35.8 (C-1,12), 34.9 (C-3,10), 34.3 (C(CH₃)₃), 31.2 (C(CH₃)₃), 17.0 (C-20); IR ν (KBr) 3040, 2962, 2214, 1592, 1478, 1225, 1185, 871, 746 cm⁻¹; UV (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 292 (2320), 311 (1210); EIMS m/z 367 (M⁺); HRMS calcd for $C_{22}H_{25}NS_2$ 367.1428, found 367.1419. Anal. Calcd: C, 71.89; H, 6.86; N, 3.81. Found: C, 71.56; H, 6.52; N, 3.70.

For synthetic purposes, both isomers could be used together in the next step.

1,10-Bis(methylthio)-5-tert-butyl-16-cyano-8-methyl-anti-[2,2] metacyclophane (75). BuLi (2.5 mmol, 1 mL, 2.5 M in hexanes) was added dropwise to a solution of either the syn- thiacyclophane 73 or anti-74 or a mixture of both (250 mg, 0.68 mmol) in dry THF (30 mL) at 0 °C. The solution was allowed to warm to room temperature over 10 min. Excess MeI (0.4 mL) was then added and the mixture allowed to stir for another 10 min, when it was quenched using water and then extracted with dichloromethane. The organic extracts were dried and evaporated. The residue was chromatographed over silica gel using CH_2Cl_2 −hexanes (12:88) as eluent to afford 242 mg (90%) of 75 as a mixture of isomers, which could be used in the next synthetic step. For characterization purposes, rechromatography yielded a single isomer of 75 in which the 1,10-methylthio groups are pseudoequatorial: mp 139−140 °C; ¹H NMR δ 7.97 (d, J = 7.7, 2H, H-12,14), 7.39 (t, J = 7.7, 1H, H-13), 7.28 (s, 2H, H-4,6), 4.09 (dd, $J = 11.2, 4.2$ Hz, 2H, H-1,10), 2.67 (dd, $J = 12.6, 4.2, 2H, H-2_{eq}, 9_{eq}$), 2.67 (dd, J = 11.4, 12.6 Hz, 2H, H-2_{ax}9_{ax}), 2.13 (s, 6H, S-Me), 1.33 (s, 9H, C(CH₃)₃), 0.54 (s, 3H, Me); ¹³C NMR δ 150.2 (C-5), 140.9 (C-8), 140.3 (C-11,15), 134.9 (C-3,7), 130.8 (C-13), 126.6 (C-12,14), 126.4 (C-4,6), 119.6 (C-16), 114.4 (CN), 54.8 (C-1,10), 42.8 (C-2,9), 34.2 $(C(CH_3)_3, 31.3 (C(CH_3)_3), 15.5 (S-Me), 14.5 (C-17); IR \nu$ (KBr) 3049, 2951, 2864, 2216, 1598, 1479, 1455, 866, 792, 743, 578 cm⁻¹; UV (cyclohexane) λ_{max} nm (ε_{max}) 274 (3040), 295 (3530); EIMS m/z 395(M⁺); HRMS calcd for $C_{24}H_{29}NS_2$ 395.1741, found 395.1747. Anal. Calcd: C, 72.86; H, 7.39; N, 3.54. Found: C, 72.41; H, 7.02; N, 3.50.

Bis-sulfonium salt 76. Using the same procedure as for 33 mixed isomers of cyclophane 75 (2.8 g, 7.0 mmol) and Borch reagent (4.5 g, 22 mmol) in dichloromethane (25 mL) gave 3.24 g (77%) of the air sensitive bis-sulfonium salt 76: ¹H NMR (DMSO- d_6) δ 8.02 (d, J = 7.9 Hz), 7.91−7.85 (m), 7.75−7.62 (m), 7.49, 7.50 (singlets), 7.35−7.27 (m), 7.22−7.12 (m), 7.08 (t, J = 6.5 Hz), 6.63, 6.60 (s), 5.84−5.54 (m), 4.85−4.74 (m), 4.73−4.67 (m), 3.45−3.30 (m), 3.10−2.92 (m), 1.34−1.30 (singlets), 1.11, 1.04, 0.97, 0.69, 0.63, 0.56 (s). These were used directly in the next step.

5-tert-Butyl-16-cyano-8-methyl-anti-[2.2]metacyclophane-1,9 diene (77). Using the same procedure as for 12, reaction of t-BuOK $(0.87 \text{ g } 7.7 \text{ mmol})$ and bis-sulfonium salt 76 $(2.0 \text{ g } 3.35 \text{ mmol})$ in THF (35 mL) followed by column chromatography using CH_2Cl_2 − hexanes (4:6) as eluent gave 710 mg (70%) of 77 as colorless crystals from cyclohexane: ¹H NMR δ 7.28 (t, J = 7.5 Hz, 1H, H-13), 6.85 (d, $J = 7.5$ Hz, 2H, H-12,14), 6.79 (s, 2H, H-4,6), 6.62 (d, $J = 11.3$ Hz, 2H, H-2,9), 6.30 (d, J = 11.3 Hz, 2H, H-1,10), 1.57 (s, 3H, Me), 1.31 $(s, 9H, -C(CH_3)$ ₃); ¹³C NMR δ 154.0 (C-5), 141.5 (C-8), 141.0 (C-11,15), 137.7 (C-2,9), 136.2 (C-3,7), 133.3 (C-13), 123.0 (C-1,10), 126.6 (C-12,14), 124.6 (C-4,6), 120.0 (C-16), 115.9 (CN), 34.5 $(-C(CH_3)_3)$, 31.5 $(-C(CH_3)_3)$, 19.8 $(C-18)$; IR ν (KBr) 3048, 3009, 2964, 2904, 2867, 2215, 1571, 1476, 1443, 1217, 1154, 873, 843, 808, 768, 679, 660, 583, 556, 490 cm $^{-1}$; UV (cyclohexane) λ_{\max} nm (ε_{\max}) 225 (31100), 287 (12440), 337 (14970), 375 (6800); EIMS m/z 299 (M^{\dagger}) ; HRMS calcd for $C_{22}H_{21}N$ 299.1674, found 299.1669. Attempted mp determination converted CPD 77 into DHP 78 and the migration product 94.

2-tert-Butyl-10c-cyano-10b-methyl-trans-10b,10c-dihydropyrene (78). Irradiation of an NMR sample of cyclophanediene 77 (10 mg) in benzene- d_6 with a UV source with output 254 nm quickly converts 77 in to 78 in ∼75% yield (∼7.5 mg) as a dark green solid: ^1H NMR δ 8.56 (s, 2H, H-1,3), 8.44 and 8.41 (AB, $J = 7.9$ Hz, 4H, H-4,5,9,10), 8.30 (d, J = 7.8 Hz, 2H, H-6,8), 7.75 (t, J = 7.8 Hz, H-7), 1.49 (s, 9H, $-C(CH_3)_3$, t-Bu), -3.97 (s, 3H, Me); ¹³C NMR δ 150.0 (C-2), 140.0 (C-10a,3a), 126.3 (C-5,9), 125.7 (C-5a,10d), 124.7 (C-6,8), 124.5 (C-4,10), 122.8 (C-7), 122.5 (C-1,3), 112.7 (CN), 36.6 $(-C(CH_3)_3)$, 33.2 (C-10c), 32.0 $(-C(CH_3)_3)$, 30.2 (C-10b), 12.0 (C-12); IR ν (thin film) 3040, 2964, 2867, 2217, 1601, 1576, 1478, 1463, 1231, 965, 878, 843, 825, 757, 736, 716 cm⁻¹; UV-vis (cyclohexane) λ_{\max} nm (ε_{\max}) 203 (50300), 274 (7500), 339 (43600), 377 (5600), 463 (3340), 533 (40), 575 (105), 600 (50). Attempted mp determination converted DHP 78 into the migration product 94. 2-tert-Butyl-10a-cyano-10b-methyl-trans-10a,10b-dihydropyr-

ene (94).

Cyclophanediene 77 (10 mg) was sealed in a glass tube under argon and heated at 70 $^{\circ}$ C for 3 h (or until completely colorless), which quantitatively converted it into colorless 94. Evaporation gave crystals: mp 82−84 °C; ¹ H NMR δ 7.18 (dd, J = 7.6, 1H, H-7), 7.07−7.03 (m, 2H, H-6,8), 6.72 (d, J = 9.1, 1H, H-9), 6.50 and 6.34 (AB, J = 9.5, 1H, H-4,5), 6.13 (s, 1H, H-3), 5.99 (d, J = 9.1 Hz, 1H, H-10), 5.53 (s, 1H, H-1), 1.157 (s, 3H, Me), 1.153 (s, 9H, $-C(CH_3)_3$); ¹³C NMR δ 150.6 (C-2), 143.2 (C-3a), 136.2 (C-10c), 131.7 (C-5a), 130.9 (C-10d), 129.0 (C-5), 127.8 (C-7), 127.4 (C-6), 126.8 (C-8), 126.6 (C-4), 125.7 (C-10), 121.52 (CN), 121.48 (C-3), 113.7 (C-1), 43.2 (C-10b), 40.8 (C-10a), 34.7 (-C(CH₃)₃), 28.9 (-C(CH₃)₃), 17.4 (C-12); IR ν (KBr) 3038, 2961, 2904, 2867, 2217, 1625, 1577, 1477, 1465, 1367, 1210, 966, 865, 825, 813, 756, 746, 661 cm[−]¹ ; UV (dichloromethane) λ_{max} nm (ε_{max}) 259 nm (12200), 273 (8870), 289 (4380), 301 (4520), 315 (3870), 339 (3660), 353 (3750); EIMS m/z 299 (M⁺) HRMS calcd for $C_{22}H_{21}N$ 299.1674, found 299.1674.

6-tert-Butyl-9-methyl-18-phenylethynyl-2,11-dithia[3.3] metacyclophanes (80) and (81). A solution of bis-bromomethyl-2 phenylethynylbenzene¹³ 79 (3.90 g, 10.7 mmol) and 2,6 bis-mercaptomethyl-4-tert-butyltoluene¹² $72(2.57 g, 10.7 mmol)$ in deaerated benzene (500 mL) wa[s a](#page-14-0)dded dropwise to an ethanolic KOH solution, prepared by adding KOH (3.4 [g](#page-14-0)) to deaerated water (160 mL) and ethanol (1470 mL) followed by addition of sodium borohydride (0.64 g). The drop rate was very crucial, and there was a drop every 5−6 s, such that the addition took ∼56 h. The solvent was then evaporated, and the residue was dissolved in dichloromethane, washed with water, dried, and evaporated. The residue was chromatographed on silica gel using CH_2Cl_2 -hexanes (15:85) as eluent. Eluted first was 700 mg (15%) of anti 81 as colorless crystals from cyclohexane: mp 148−149 °C; ¹H NMR δ 7.44–7.39 (m, 2H, H-22, 26), 7.40 (d, J = 7.6 Hz, 2H, H-14,16), 7.33−7.31 (m, 3H, H-23,24,25), 7.27 (s, 2H, H-5,7), 7.22 (t, $J = 7.6$ Hz, 1H, H-15), 4.16 and 3.72 (AB, $J = 14.1$ Hz, 4H, H-1, 12), 3.80−3.70 (m, 4H, H-3,10), 1.42 (s, 3H, Me), 0.98 (s, 9H, [−]C(CH3)3.); 13C NMR ^δ 148.3 (C-6), 139.3 (C-13,17), 135.9 (C-9), 134.1 (C-4,8), 131.6 (C-22,26), 129.1 (C-14,16), 128.51 (C-23,25), 128.45 (C-24), 127.8 (C-15), 127.3 (C-5,7), 125.7 (C-18), 124.0 (C-21), 100.2 $(C-20)$, 86.2 $(C-19)$, 34.1 $(-C(CH_3)_3)$, 32.4 $(C-3,10)$, 32.1 $(C-1,12)$, 31.1($-C(CH_3)$), 15.3 (C-27); IR ν (KBr) 3057, 2959, 2904, 2862, 1594, 1477, 1488,1440, 756. 7477, 691 cm⁻¹; UV (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 288 (10100) 303 (11700), 318 (11900); EIMS m/z 442 (M⁺); HRMS calcd for $C_{29}H_{30}S_2$ 442.1788, found 442.1779.

Eluted second was 2.20 g $(47%)$ of syn-80 as colorless crystals from cyclohexane: mp 170−171 °C; ¹H NMR δ 7.62 (dd, J = 7.8, 1.8 Hz, 2H, H-22,26), 7.40−7.44 (m, 3H, H-23,24,25), 7.02 (d, J = 7.6, 2H, H-14,16), 7.01 (s, 2H, H-5,7), 6.86 (t, J = 7.6 Hz, 1H, H-15), 4.65 and 3.67 (AB, $J = 14.8$ Hz, 4H, H-1,12), 4.41 and 3.62 (AB, $J = 14.8$ Hz, 4H, H-3,10), 2.54 (s, 3H, Me) 1.22 (s, 9H, $-C(CH_3)_3$); ¹³C NMR δ 147.2 (C-6), 140.4 (C-13,17), 136.0 (C-4,8), 132.2 (C-9), 131.4 (C-22,26), 129.0 (C-15), 128.84 (C-23,25), 128.75 (C-24), 128.0 (C-14,16), 126.0 (C-5,7), 123.6 (C-21), 121.7 (C-18), 99.2 (C-20), 87.8 (C-19), 35.8 (C-1,12), 34.7 (C-3,10), 34.3 (-C(CH₃)₃), 31.3

 $(-C(CH_3)$ ₃), 17.2 (C-27); IR ν (thin film, KBr) 3057, 2961, 2906, 2864, 1596, 1489, 1478, 1461, 1361, 1261, 872, 792, 755, 748, 691 cm[−]¹ ; UV (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 258 (16100), 297 (13600), 319 (11000); EIMS m/z 442 (M⁺); HRMS calcd for C₂₉H₃₀S₂ 442.1788, found 442.1786.

There was no need to separate 80and 81 for synthetic purposes.

1,10-Bis(methylthio)-5-tert-butyl-8-methyl-16-phenylethynylanti-[2,2]metacyclophane (82). BuLi (2.5 mmol, 1 mL, 2.5 M in hexanes) was added dropwise to a solution of mixed thiacyclophanes 80/81 (300 mg, 0.68 mmol) in dry THF (30 mL) at 0 °C. The solution was allowed to warm up to room temperature over 10 min when the dark solution had turned pale yellow. Excess MeI (0.4 mL) was added and was allowed to stir for another 10 min. The reaction was quenched with water (5 mL) and then was extracted with dichloromethane. The organic extracts were dried and evaporated. The residue was chromatographed over silica gel using $CH₂Cl₂$ -hexanes (12:88) as eluent and gave 290 mg (80%) of mixed isomers of 82. For characterization purposes, rechromatography yielded 150 mg of a single isomer of 82 in which the 1,10 methylthio groups are pseudoequatorial: mp 203−204 °C; ¹H NMR δ 7.86 (d, J = 7.6 Hz, 2H, H-12,14), 7.28 (tt, J = 6.4, 1.5 Hz, 1H, H-22), 7.27−7.17 (m, 4H, H-20,21,23,24), 7.21 (t, $J = 7.6$ Hz, 1H, H-13), 7.14 (s, 2H, H-4,6), 4.45 (dd, J = 11.4, 4.1 Hz, 2H, H- 1_{ax} 10_{ax}), 3.24 (dd, J = 12.4, 4.1 Hz, 2H, H-2_{eq},9_{eq}), 2.74 (dd, J = 12.2, 11.6 Hz, 2H, H-2_{ax},9_{ax}), 2.18 (s, 6H, S-Me), 0.91 (s, 9H, $-C(CH_3)_3$), 0.63 (s, 3H, Me); ¹³C NMR δ 148.4 (C-5), 140.0 (C-8), 138.8 (C-11,15), 135.2 (C-3,7), 131.7 (C-13), 130.2 (C-16), 128.3 (C-22), 127.3, 128.5 (C-20,21,23,24), 125.8 (C-12,14), 125.2 (C-4,6), 123.6 (C-19), 92.8 (C-18), 86.2 (C-17), 54.6 (C-1,10), 43.5 (C-2,9), 34.1 (-C(CH₃)₃), 31.3 (-C(CH₃)₃), 15.9 (C-30,31), 14.9 (C-25); IR (thin film, KBr) 3044, 2952, 2913, 1596, 1490, 1425, 754, 745, 690 cm⁻¹; UV (cyclohexane) λ_{max} nm $(\varepsilon_{\text{max}})$ 280 (13000), 293 (13500), 320 (12600); EIMS m/z 470 (M+); HRMS calcd for $C_{31}H_{34}S_2$ 470.2102, found 470.2104. Anal. Calcd: C, 79.09; H, 7.28. Found: C, 79.35; H, 7.01

1, 10-Bis(methylthio)-8-methyl-16-phenylethynyl-anti-[2.2] metacyclophane Bis-sulfonium Salt (83). Using the same procedure as for 33, the pure 1,10-diequatorial-methylthio isomer of 83 (1.0 g, 2.1 mmol) on reaction with Borch reagent (1.0 g, 4.8 mmol) in CH_2Cl_2 (7 mL) gave 1.4 g (quant.) of 83 as a single isomer: ¹H NMR $(DMSO-d₆)$ δ 7.87 (d, J = 7.8 Hz, 2H, H-12,14), 7.58 (dd J = 7.8, 2.3 Hz, 2H, H-20,24), 7.47 (t, J = 7.76 Hz, 1H, H-13), 7.41 (m, 3H, H-21−23), 7.35 (s, 2H, H-4,6), 4.66 (dd, J = 12.0, 4.2 Hz, 2H, H-1_{ax},10_{ax}), 3.60 (dd, J = 12, 4.2 Hz, 2H, H-2_{eq},9_{eq}), 3.46 (s, 6H, S-Me₂), 3.17 (triplet, $J = 12$ Hz, 2H, H-2_{ax}, 9_{ax}), 3.04 (s, 6H, S-Me₂), 0.89 (s, 9H, $-C(CH_3)$,), 0.75 (s, 3H, Me). When mixed isomers were used the following NMR data was obtained: ^1H NMR (DMSO- d_6) δ 7.88−7.34 (m), 7.0−6.8 (m), 6.6 (br s), 5.83−5.55 (m), 4.8−4.6 (m), 4.04−3.00 (m), 2.26, 1.98 (s), 1.18, 0.78 (9 singlets). These were used in the next step.

5-tert-Butyl-8-methyl-16-phenylethynyl-anti-[2.2]metacyclophane-1,9-diene (84). Using the same procedure as for 12, reaction of bissulfonium salt 83 (300 mg, 0.45 mmol) and t -BuOK (120 mg, 1.1) mmol) in THF (10 mL) followed by chromatography over silica gel in a foil wrapped column using hexanes/ CH_2Cl_2 (80:20) as eluent gave 147 mg (90%) of CPD 84 as colorless crystals: ¹H NMR δ 7.25-7.23 $(m, 5H)$, 7.12 $(t, J = 7.4 \text{ Hz}, 1H)$, 6.77 $(d, J = 7.4 \text{ Hz}, 2H)$, 6.58 $(s,$ 2H), 6.49 (d, J = 11.3 Hz, 2H), 6.36 (d J = 11.3 Hz, 2H), 1.65 (s, 3H, Me), 0.80 (s, 9H, $C(CH_3)_3$, ^tBu); ¹³C NMR δ 150.6, 141.7, 137.4, 131.3, 130.2, 129.4, 128.3, 127.8, 125.9, 124.6, 123.1, 95.1, and 90.0 $(C\equiv C)$, 33.0 (C-26), 30.7 (-C $(CH_3)_3$) 20.2 (Me). ¹H NMR (C₆D₆) δ 7.28 (dd, J = 7.1, 1.5 Hz, 2H, H-20,24), 6.92−6.98 (m, 3H, H-21−23), 6.86 (t, J = 7.4 Hz, 1H, H-13), 6.73 (s, 2H, H-4,6), 6.54 (d J = 7.4 Hz, 2H, H-12,14), 6.42 and 6.23 (AB, J = 11.3 Hz, 2H, H-1,2,9,10), 1.83 (s, 3H, Me), 0.93 (s, 9H, $-C(CH_3)_3$); ¹³C NMR (C₆D₆) δ 150.9 (C-5), 142.5 (C-8), 141.2 (C-11,15), 138.1 (C-3,7), 135.4 (C-2,9), 132.3 (C-20,24), 132.0 (C-1,10), 131.1 (C-16), 129.8 (C-13), 128.7 (C-21,23), 128.1 (C-22), 126.3 (C-12,14), 125.5 (C-19), 123.7 $(C.4,6)$, 95.9 (C-18), 91.0 (C-17), 34.2 (-C(CH₃)₃), 31.7 (-C(CH₃)₃), 20.8 (C-25); IR ν (thin film) 3041, 3004, 2867, 2962, 1597, 1491, 1442, 871, 772, 757, 737, 690, 673 cm⁻¹; UV (cyclohexane) λ_{max} nm

 $(\varepsilon_{\text{max}})$ 288 nm (13500), 339 (15000), 377 (6700); EIMS 374 (M⁺). Attempted mp determination converted CPD 84 into DHP 85.

2-tert-Butyl-10b-methyl-10c-phenylethynyl-trans-10b,10c-dihydropyrene (85). t-BuOK (140 mg, 1.2 mmol) was added to a stirred suspension of mixed isomers of bis-sulfonium salt 83 (350 mg, 0.52 mmol), in THF (10 mL) under argon at 20 °C. After refluxing for 6 h, water was added and then dichloromethane (90 mL). The extract was washed, dried, and evaporated. The residue was chromatographed over silica gel using hexanes/CH₂Cl₂ (85:15) as eluent to give 160 mg (85%) of the dihydropyrene 85 as green crystals from methanol: mp 112−113 °C; ¹H NMR δ 8.72 and 8.65 (AB, J = 7.8 Hz, 2H, H-4,5,9,10), 8.69 (s, 2H, H-1,3), 8.62 (d, J = 7.8 Hz, 2H, H-6,8), 8.07 $(t, J = 7.8 \text{ Hz}, 1H, H-7)$, 6.83 (tt, $J = 7.8$, 1.8 Hz, 1H, H-16), 6.72 (dd, J = 7.8, 7.5 Hz, 2H, H-15,17), 6.04 (dd, J = 7.4, 1.8 Hz, 2H, H-14,18), 1.74 (s, 9H, $-C(CH_3)_3$), -4.00 (3H, s, Me); ¹³C NMR δ 148.2 (C-2), 140.0 (C-3a,10a), 131.1 (C-14,18), 130.1 (C-5a,10d), 127.4 (C-15,17), 127.0 (C-16), 125.1 (C-5,9), 123.5 (C-4,10), 123.3 (C-6,8), 122.5 (C-13), 122.0 (C-7), 121.3 (C-1,3), 84.6 (C-12) 75.4 (C-11), 36.4 $\left(\text{-}C(CH_3)_3\right)$, 32.1 $\left(\text{-}C(CH_3)_3\right)$, 31.2 $\left(\text{C-10b}\right)$, 30.5 $\left(\text{C-10c}\right)$, 12.9 (C-19); IR ν (thin film) 3036, 2963, 1597, 1489, 1478, 1442, 1230, 874, 822, 755, 737, 691, 682, 628 cm⁻¹; UV–vis (cyclohexane) λ_{\max} nm $(\varepsilon_{\text{max}})$ 250 (41200), 339 (1.64 \times 10⁵), 377 (73200), 467 (14900), 577 (380), 639 (300); EIMS 374 (M⁺); HRMS calcd for $C_{29}H_{26}$ 374.2034, found 374.2036.

8,16-Diformyl-anti-[2.2]metacyclophane-1,9-diene (27). DIBAL (2.5 mmol, 2.5 mL of 1 M solution in cyclohexane) was added dropwise to a solution of dicyano CPD³ 5 (250 mg, 1 mmol) in dry dichloromethane (12 mL) at room temperature, over a period of 6 min. It was stirred for 5 h at 20 °C, after [wh](#page-14-0)ich it was slowly added to methanol (5 mL) and stirred for 30 min. Then HCl (10 mL, 1M) was added carefully and the resulting solution was extracted with dichloromethane . The organic layer was dried, evaporated, and column chromatographed over silica using ethyl acetate:dichloromethane (1:9) as eluent to yield 146 mg (56%) of the diformyl-CPD 27 as colorless crystals from methanol: ^{1}H NMR δ 9.35 (s, 2H, H-17,18), 7.39 (t, J = 7.4 Hz, 2H, H-5,13), 6.85 (d, J = 7.4 Hz, 4H, H-4,6,12,14), 6.73 (s, 4H, H-1,2,9,10); ¹³C NMR) δ 185.8 (C-17,18), 145.4(C-8,16), 140.0 (C-3,7,11,15), 135.4 (C-1,2,9,10), 135.3 (C-5,13), 128.8 (C-4,6,12,14); IR ν (KBr) 3056, 3011, 2924, 2869, 2768, 1712, 1683, 1559, 1445, 1222, 1181, 816, 761, 628 cm⁻¹; UV-vis (dichloromethane), λ_{max} nm (ε_{max}) 245 nm (8500), 274 (8600), 338 (4150), 345 (4130), 395 sh (1500); EIMS m/z 260 (M⁺). Attempted mp determination converted CPD 27 into migration product 30.

1,10a-Diformyl-trans-1,10a-dihydropyrene (30). Diformyl-CPD 27 (10 mg) in CDCl₃ (1 mL) was sealed in a glass tube under argon and heated at 70 °C for 1 h, which converted it into 30. Evaporation gave colorless solid (10 mg, 100%) which decomposed on attempted chromatography but was pure enough to obtain the following data: ¹H NMR δ 9.71 (d, J = 3.4 Hz, 1H, H-18), 9.34 (s, 1H, H-17), 7.69 (d, J = 8.3 Hz, 1H, H-5), 7.61 (d, J = 8.4 Hz, 1H, H-6), 7.32 (dd, J = 6.9, 8.4 Hz, 1H, H-7), 7.28 (d, J = 8.3 Hz, 1H, H-4), 7.11 (d, J = 6.8 Hz, 1H, H-8), 6.9 (d, J = 9.6 Hz, 1H, H-9), 6.7 (d, J = 9.9 Hz, 1H, H-3), 5.91 $(dd, J = 9.6, 6.2 Hz, 1H, H-2), 5.68 (d, J = 9.6 Hz, 1H, H-10), 3.76$ \dot{d} dd, \dot{J} = 6.3, 3.5 Hz, 1H, H-1); ¹³C NMR δ 198.7 (C-18), 191.0(C-17), 133.5 (C-5a), 132.9 (C-9), 131.3 (C-3), 131.2 (C-3a), 130.5 (C-10d), 128.6 (C-6), 128.41 (C-10c), 128.2 (C-5), 127.1 (C-7), 126.1 (C-4), 125.6 (C-10b), 125.3 (C-8), 122.7 (C-10), 122.0 (C-2), 55.9 (C-10a), 51.1 (C-1); IR ν (thin film) 3048, 2927, 2820, 2721, 1719, 1685, 1590, 1569, 1502, 1174, 842, 770, 734 cm[−]¹ ; UV−vis (dichloromethane) λ_{max} nm $(\varepsilon_{\text{max}})$ 208 (850), 243 (13500), 274 (24600), 322 (3600), 322 (3600), 338 (4100), 402 (2130), 470 sh (460); EIMS m/z 260 (M⁺)

Attempted chromatography of 30 led to formation of 1-formylpyrene 31.

■ ASSOCIATED CONTENT

S Supporting Information

Table S1: Thermodynamic parameters for the thermal back reaction: cyclophanediene to dihydropyrene. UV−vis spectrum of the orange plastic cutoff filter used for the photo-opening

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experiments. General synthetic experimental conditions. The numbering system used for NMR assignments. ${}^{1}H$ and ${}^{13}C$ NMR spectra for all new compounds. Thermal closing data with Arrhenius and Eyring plots for the thermal closing reaction for the compounds studied. This material is available free of charge via the Internet at http://pubs.acs.org.

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